

Chapter 12

SECTION 3:

Problem: Answer the following questions in the space provided.

1. Illustration for the pressure of a fixed mass of gas with constant volume, at two different in the following conditions:

- Condition: a temperature increase, volume kept constant
- Condition: a volume increase, temperature kept the same
- Condition: a temperature decrease, a volume kept the same
- Condition: a volume decrease, temperature kept the same

2. Two sealed flasks, A and B, contain the different gases at equal volume at the same temperature conditions. Assume that flask A is connected to flask B by a tube. With the pressure in the two flasks remains equal? If not, which flask will have the higher pressure? Will the pressure will not increase equal. If all other factors remain constant flask A will have higher pressure because it is at a higher temperature.

Problem: Answer the following questions in the space provided.

3. A large tire is inflated to 30 psi at 30°C. Assume that the volume of the tire stays constant approximately when the tire is inflated.

a. If the tire is inflated the air inside is at 30°C by 30 psi. Assume that the pressure in the tire is constant at 30 psi. Assume the volume of the tire is constant. Assume the temperature is constant at 30°C.

b. Assume the temperature is 30°C. Assume the pressure is 30 psi. Assume the volume is constant at 30°C.

c. Assume the pressure is 30 psi. Assume the temperature is 30°C. Assume the volume is constant at 30°C.

Review Gases Section 2 Short Answer

**Mikhail V. Kurushkin, W. H. Eugen
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