Name:				_ Period:
1. What is a va	lence electron and w	hy are they import	ant to a chemist?	
Valence electri	ons are used in bond	ing and determine	properties/character	istics of elements
2. How can you	find the number of	valence electrons	an element has?	
The group num	ber on the periodic t	table tells you the	number of valence ele	ectrons an element k
3. How many pr	otons are in the foll	lowing elements?		
A. Carbon	B. Hydrogen	C. Oxygen	D. Phosphorus	E. Calcium
6	1	8	15	20
4. Draw the at	omic structures for	the following. Inclu	de the number of pr	otons, neutrons and
draw in the val	ence electrons.	_	-	
A. Silicon	Silicon B. Potassium			
Silicon				
39			_	
12:31				
(⊕);;			((((())))	
C. Sulfur			D. Beryllium	
Swiffur				
3			-	
(60)			((\varTheta))	
•				
E. Argon			F. Nitrogen	
			Nitrogen	
	Les .			

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