Use the information in these tables to complete the graph as described below

Symbol	Atomic Radius (Picometers)	First Ionization energy (kilojoules/mole)	(4-point scale)	
H	31	1312	2.1	
I., i	128	520	1.0	
Nia	166	496	0.9	
K	203	410	0.8	
Rb	220	403	0.8	
Cs	244	376	0.7	

Symbol	Atomic Radius (Picometers)	First lonization energy (kilojoules/mole)	Electronegativity (4-point scale) 0.9 1.2	
Na	166	496		
Mg	141	738		
AL	121	578	1.5	
Sii	111	787	1.8	
P	107	1012	2.1	
S	105	1000	2.5	
CL	102	1251	3.0	
Ar	106	1521		

- Using colored pencils or pen, list under the symbol (in this order!) the Atomic Radius, First Ionization Energy, and Electronegativity.
- Use a different color for each property. Example: Write all of the Atomic radius values in red, all of the First
 ionization energies in green, and all of the Electronegativities in blue. Of course, you can pick any colors that are
 available, as long as you are consistent.
- 3. Observe the trends in each property as you go down the Alkali metal group, and as you go across Period 3.
- Write out each of the statements written on the opposite side, completing each statement with the observed trend (increase or decrease)

Li	Across a period (→) electronegativity tends to						
Na 166 496 0.9	Mg	Al	Si	Р	s	CI	A
к	Down	a Group (atomic rac	lius tends t	٥		
Rb	0.0000000000000000000000000000000000000		first ioniza		tends to _		

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Henry Bent

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The Periodic Table Eric R. Scerri, 2006-10-12 The periodic table is one of the most potent icons in science It lies at the core of chemistry and embodies the most fundamental principles of the field. The one definitive text on the development of the periodic table by van Spronsen 1969 has been out of print for a considerable time. The present book provides a successor to van Spronsen but goes further in giving an evaluation of the extent to which modern physics has or has not explained the periodic system The book is written in a lively style to appeal to experts and interested lay persons alike The Periodic Table begins with an overview of the importance of the periodic table and of the elements and it examines the manner in which the term element has been interpreted by chemists and philosophers. The book then turns to a systematic account of the early developments that led to the classification of the elements including the work of Lavoisier Boyle and Dalton and Cannizzaro The precursors to the periodic system like D bereiner and Gmelin are discussed In chapter 3 the discovery of the periodic system by six independent scientists is examined in detail Two chapters are devoted to the discoveries of Mendeleev the leading discoverer including his predictions of new elements and his accommodation of already existing elements Chapters 6 and 7 consider the impact of physics including the discoveries of radioactivity and isotopy and successive theories of the electron including Bohr's quantum theoretical approach Chapter 8 discusses the response to the new physical theories by chemists such as Lewis and Bury who were able to draw on detailed chemical knowledge to correct some of the early electronic configurations published by Bohr and others Chapter 9 provides a critical analysis of the extent to which modern quantum mechanics is or is not able to explain the periodic system from first principles Finally chapter 10 considers the way that the elements evolved following the Big Bang and in the interior of stars The book closes with an examination of further

chemical aspects including lesser known trends within the periodic system such as the knight's move relationship and secondary periodicity as well at attempts to explain such trends Science Comics: The Periodic Table of Elements Ion Chad, 2023-03-14 Step out of your element with Science Comics The Periodic Table of Elements the latest volume of First Second's nonfiction graphic novel series A book of fun chemistry experiments has fallen into the wrong hands Only Mel can use her knowledge of the periodic table to put an end to a maniacal madman s evil schemes The periodic table helps us quickly understand the 118 elements those tiny substances that make up everything in the world By using the periodic table we can recognize how these building blocks behave find trends and patterns in the universe and make predictions about elements that haven t been discovered yet Join us in learning about the periodic table and maybe the next big discovery will **Trends in the Periodic Table** Open University. S25- Course Team,1972 Periodic Table: A Formula **Handbook** N.B. Singh, Periodic Table A Formula Handbook is a concise and indispensable guide to the elements providing a comprehensive collection of essential formulas properties and trends within the periodic table This handbook equips students scientists and enthusiasts with quick access to vital information on each element including atomic number atomic mass electron configuration and chemical properties With clear organization and easy to understand explanations this book serves as an invaluable resource for anyone studying chemistry conducting research or simply seeking to deepen their understanding of the fundamental building blocks of matter **Chemical Periodicity** Robert Thomas Atoms, Molecules & Elements: Patterns In the Periodic Table Gr. 5-8 George Sanderson, 2013-04-20 Graybill, 2015-10-01 This is the chapter slice Patterns In the Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives TRENDS IN THE PERIODIC TABLE OF THE **ELEMENTS COMPUTER BASED INSTRUCTION IN CHEMISTRY.**, 1992 Atoms, Molecules & Elements: The **Periodic Table Gr. 5-8** George Graybill, 2015-10-01 This is the chapter slice The Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of

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