

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____

Graphing Periodic Table Trends

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	152	Sodium - 11	190	Potassium - 19	243	Rubidium - 37	265
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 38	219
Boron - 5	87	Aluminum - 13	118	Gallium - 31	136	Indium - 49	156
Carbon - 6	67	Silicon - 14	111	Germanium - 32	125	Tin - 50	145
Nitrogen - 7	56	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	115

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 37	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 38	550

Atomic Radius



Trends In The Periodic Table Answers Graphing

Randy L. Bell, Joe Garofalo



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Chemistry in the Community. American Chemical Society, 2002 This volume has relevance to a wide number of courses giving a hands on introduction to chemistry in relation to community issues rather than around specific chemical concepts. *Chemistry in the Community (ChemCom)* American Chemical Society, 2011-06-17 Touted as the most successful NSF funded project published *Chemistry in the Community ChemCom* by the American Chemical Society ACS offers a meaningful and memorable chemistry program for all levels of high school students. ChemCom covers traditional chemistry topics within the context of societal issues and real world scenarios. Centered on decision making activities where students are responsible for generating data in an investigation, analyzing that data and then applying their chemistry knowledge to solve the presented problem. The text is intensively laboratory based with all 39 of the investigations integrated within the text, not separate from the reading. With the ChemCom program students learn more organic and biochemistry, more environmental and industrial chemistry and more on the particulate nature of matter than other textbooks. All within the relevance of solving problems that arise in everyday life. Meticulously updated to meet the needs of today's teachers and students, the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards. Incorporating advances in learning and cognitive sciences, ChemCom's wide ranging coverage builds upon the concepts and principles found in the National Science Education Standards. Correlations are available showing how closely aligned ChemCom is to these and other state standards. *Science Units for Grades 9-12* Randy L. Bell, Joe Garofalo, 2005 Sample topics include cell division, virtual dissection, earthquake modeling, the Doppler Effect and more. *Jacaranda Chemistry 1 VCE Units 1 and 2, learnON and Print* Neale Taylor, Angela Stubbs, Robert Stokes, 2022-11-30 Developed by expert Victorian teachers for VCE students. The NEW Jacaranda Chemistry VCE series continues to deliver curriculum aligned material that caters to students of all abilities. Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design 2023-2027. *Principles of*

Epidemiology Workbook: Exercises and Activities Ray M. Merrill, 2010-09-15 This workbook was written for students of epidemiology and serves as a supplement to any one of several introductory text books in epidemiology Each chapter is divided into an introduction a series of questions and detailed responses and a series of Homework questions At the end of each chapter is a table with a list of selected epidemiology text books with accompanying chapters in those books that the workbook chapter may supplement The general learning outcomes LOs for this workbook are 1 Become familiar with basic concepts and definitions commonly used in epidemiology 2 Define a public health problem 3 Identify appropriate uses and limitations of data and research design strategies for solving public health problems 4 Make relevant inferences from quantitative and qualitative data 5 Distinguish between statistical association and cause effect relationships 6 Measure and describe patterns of disease incidence prevalence and mortality 7 Identify environmental factors and behaviors associated with health related states or events 8 Be familiar with the steps for investigating disease outbreaks 9 Identify calculate and interpret common indices used in identifying the health status 10 Evaluate program effectiveness 11 Critically assess epidemiological research 12 Be able to communicate health findings Each chapter features 10 20 mastery check questions with detailed answers 5 optional problems A case study A multiple choice short answer quiz Answers to the cases and quizzes are provided as part of the online instructor resource package

Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life

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Understanding Advanced Physical Inorganic Chemistry: The Learner's Approach (Revised Edition) Kim Seng Chan, Jeanne Tan, 2016-09-26 This revised edition has been updated to meet the minimum requirements of the new Singapore GCE A level syllabus that would be implemented in the year 2016 Nevertheless this book is also highly relevant to students who are studying chemistry for other examination boards In addition the authors have also included more Q A to help students better understand and appreciate the chemical concepts that they are mastering

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