



Nuclear Radiation Worksheet

a. What determines whether an atom is stable or unstable?

It's overall size and a difference between proper it and president it

b. How does an unstable atom gain stability?

It undergoes redisentine decay

- c. What part of an atom undergoes change during radioactive decay? The nucleus !
- 2. Which of the human senses can detect radiation?

och of the human senses can detect radiation?

Our eyes can detect colors, but we have no senses he detect alpha, but no radiotion

Explain the difference between an isotope and a radioactive isotope.
 Δε του με τ

Particle	Mass	Charge	Penetrating power
Alpha	H since I perfect	Positive (4)	Storand by process or property
Beta	O No Kees To oneway	Disease (-)	singular sheets share had or think maps zine
Gamma	O Nomer	Nechal	Should be sevel index of land or feet

5. How are the mass number and atomic number of a nucleus affected by the loss of the following?

Particle	Mass Number Change	Atomic Number Change
Alpha	-4	-2
Beta	No change	41
Gamma	No element	No charre

6. Complete the equations for the decay processes below by supplying the missing atomic numbers and mass numbers.

Example: ${}^{218}_{84}Po \rightarrow {}^{4}_{2}He + {}^{214}_{82}Pb$

Complete the following nuclear equations:

a.
$$^{239}_{83}Bi$$
 \rightarrow $^{4}_{2}He$ + $^{255}_{21}\mathcal{T}_{2}$ b. $^{239}_{93}Np$ \rightarrow $^{239}_{94}Pu$ + $^{-1}_{-1}\mathcal{C}$

Snyder 2007 Nuclear Radiation Worksheet Answer Key

Marco Cascella

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