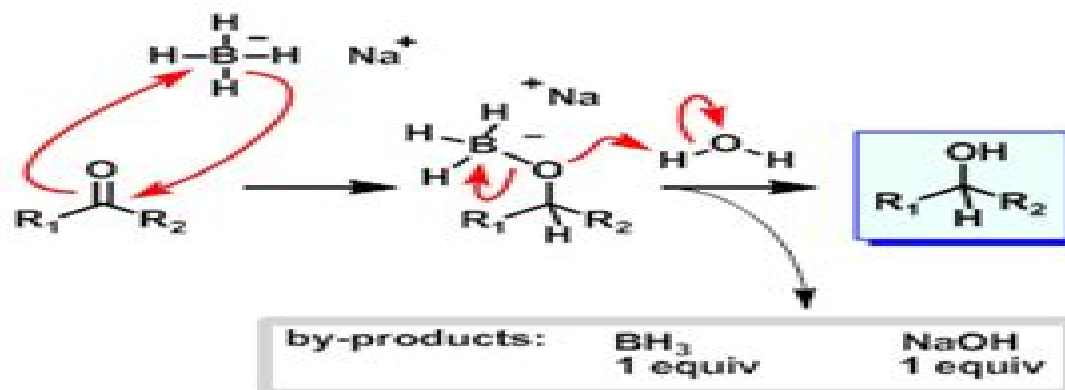


Practical 1- Borohydride reduction



Figure 3: Reduction of Benzil to hydrobenzoin with Possible Formation of Three Stereochemical Products.

- We use sodium borohydride and methanol.
- There are two steps to the mechanism:
 1. Reduction with sodium borohydride (borohydride is nucleophilic)
 2. Addition of HCl \rightarrow acid workup process where you destroy any remaining borohydride. Protonation of the basic alkoxide with hydrochloric acid to form the alcohol.
- BH_4^- - CH_4 NH_4^+ All have 8 valence electrons therefore form a tetrahedron shape
- $\text{N}(\delta^-)$ - $\text{H}(\delta^+)$. But as N is more electronegative than C, therefore $\rightarrow \text{C}(\delta^+) - \text{H}(\delta^-)$



- The phenol ring is represented on the IR spectrum which is approximately 3000cm^{-1} . Therefore, when we stretch this molecule, it leads to a large spread of charge.
- If we use NaB_2D_4 instead of NaBH_4 , we get a product which has a 2D instead of H. We can then detect this using IR.
- Use symmetry in NMR to determine the environments in protons. If the molecule does not sit on a planar - this can lead to different proton interactions, forming different environments.

Sodium Borohydride Practical Manual

Peter J. Cragg



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