

Potentiometry Questions and Answers- GROUP 13

 Bolded letter are the answers

- The electrolyte solution within the glass electrode (ref) of the pH meter is:
A. Saturated KCl
B. Concentrated HCl
C. Dilute KCl
D. Dilute HCl
- The reaction that occurs in the electrolyte solution of the pCO₂ electrode resulting in a change in pH is:
A. $\text{CO}_2 + \text{H}_2\text{O} \rightleftharpoons \text{H}_2\text{CO}_3 \rightleftharpoons \text{H}^+ + \text{HCO}_3^-$
B. $\text{H}^+ + \text{CO}_3^{2-} \rightarrow \text{H}_2\text{CO}_3$
C. $\text{H}_2\text{CO}_3 \rightarrow \text{H}_2\text{O} + \text{CO}_2$
D. $\text{CO}_2 + \text{H}_2\text{O} \rightleftharpoons 2\text{H}^+ + \text{CO}_3^{2-}$
- Which of the following is NOT associated with a pCO₂ electrode?
A. Membrane permeable to CO₂
B. pH electrode
C. Measurement is amperometric
D. Measurement is potentiometric
- Electrodes made of pH sensitive glass may be used in the analysis of:
A. PCO₂
B. PO₂
C. PH
D. All of the above
E. None of the above
- The EMF of the sample cell is determined by which of the following formulas:
A. $E_{\text{cell}} = E_{\text{cathode}} + E_{\text{Ij}} - E_{\text{anode}}$
B. $E_{\text{cell}} = E_{\text{anode}} - E_{\text{Ij}} - E_{\text{cathode}}$
C. $E_{\text{cell}} = E_{\text{cathode}} - E_{\text{Ij}} - E_{\text{anode}}$
D. $E_{\text{cell}} = E_{\text{anode}} + E_{\text{Ij}} - E_{\text{cathode}}$
- In potentiometry, which of the following is considered the standard electrode?
B. Hydrogen electrode
A. Calcium electrode
C. Potassium electrode
D. Copper electrode

Potentiometry Multiple Choice Questions

A Gutmann



Potentiometry Multiple Choice Questions:

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