

CHEM 101 Final Exam with complete solutions

How many significant figures are in this number, 0.01160 - ANSWER: → 4

Select the correct scientific notation for 117000 - ANSWER: → 1.17×10^5

the temperature of an experiment is designed at 122 degrees F. What is it in Celsius? - ANSWER: → 50 degrees Celsius

An experiment is performed at -173 degrees C. What is the temperature in Kelvin? - ANSWER: → 100K

A piece of metal has a mass of 340 g and a volume of 17.6 mL. What is the density of the metal? - ANSWER: → 19.3 g/mL

The comparison of the density of a substance with the density of water is defined as _____. - ANSWER: → specific gravity

The following forms of energy can be classified as potential energy EXCEPT _____.

A). Gasoline B). Running

C). Peanut butter sandwich D). Water held by a dam - ANSWER: → running

The following examples are chemical changes EXCEPT

A). Rusting iron B). Cooking an egg

C). Burning gasoline D). Melting ice - ANSWER: → melting ice

The amount of heat necessary to raise the temperature of 1 g liquid water by 1°C, is

defined as _____. - ANSWER: → a calorie

Possible Objective Questions On Chem 101 With Solutions

Victor M. Corman



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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Possible Objective Questions On Chem 101 With Solutions**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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