

8TH GRADE

SCIENCE EOG

REVIEW

MATTER DEFINITIONS



1. **Atom** - The smallest component of an element having the chemical properties of the element.

2. **Element** - A substance that cannot be broken down into simpler substances by chemical means.

3. **Compound** - A substance formed by the chemical combination of two or more elements in a fixed ratio.

4. **Mixture** - A substance formed by the physical combination of two or more substances in any ratio.

5. **Solution** - A homogeneous mixture of two or more substances.

6. **Suspension** - A heterogeneous mixture of two or more substances.

7. **Colloid** - A mixture of two or more substances in which the particles are intermediate in size between those in a solution and those in a suspension.

8. **Crystal** - A solid material whose atoms, molecules, or ions are arranged in an orderly repeating pattern.

9. **Amorphous** - A solid material whose atoms, molecules, or ions are not arranged in an orderly repeating pattern.

10. **Isotopes** - Atoms of the same element having the same chemical properties but different physical properties.

11. **Alloys** - A mixture of two or more metals or a metal and a non-metal.

No Matter How the Dice Fall

Roll the Dice	Write a sentence describing the event.
1	
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Swampy Words

Word	Definition	Example
Amorphous	Without a definite shape or form; irregular.	Amorphous solid
Crystalline	Having a regular, repeating arrangement of atoms or molecules.	Crystalline solid
Isotopes	Atoms of the same element having the same chemical properties but different physical properties.	Isotopes of carbon
Alloys	A mixture of two or more metals or a metal and a non-metal.	Alloys of steel
Suspensions	Heterogeneous mixtures in which the particles are large enough to be seen with the naked eye.	Suspensions of sand in water
Solutions	Homogeneous mixtures in which the particles are small enough to be dissolved.	Solutions of salt in water

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