

**CHEMISTRY 161**  
**THERMO WORKSHEET**  
**FALL 2015**

1. A system absorbs 52.5 J of energy as it expands to a volume of 50.0 L. If the external pressure is 0.800 atm and the change in internal energy is -112 J, calculate the initial volume.

2. Use Hess's law to find the heat of combustion of methane,



using the following thermochemical equations.



3. How much glucose must be metabolized to completely evaporate 1.00 g of water? The heat of combustion of glucose is 2803 kJ/mole.
4. The standard enthalpy of formation of  $\text{NH}_3$  is -46.1 kJ/mole. What is  $\Delta H^\circ_{\text{rxn}}$  for the following reactions?
- a.  $\text{N}_2(g) + 3 \text{H}_2(g) \rightarrow 2\text{NH}_3(g)$       b.  $\text{NH}_3(g) \rightarrow \text{N}_2(g) + 3 \text{H}_2(g)$
5. The heat of combustion of pentane ( $\text{C}_5\text{H}_{12}$ ) is -3535 kJ/mole
- a. Find the heat given off when 1.00 kJ of pentane is burned.
- b. How many grams of pentane must be burned to increase the temperature of 1.00 kg of water from 20.0°C to 90.0°C?
6. Brass has a density of 8.40 g/cm<sup>3</sup> and a specific heat of 0.385 J/°C. A 15.2 cm<sup>3</sup> piece of brass at an initial temperature of 163°C is dropped into an insulated container with 150.0 g H<sub>2</sub>O at 22.4°C. What will be the final temperature of the brass-water mixture?

# Thermochemistry Practice Problems Worksheet

**David E. Newton**



## Thermochemistry Practice Problems Worksheet:

*General Chemistry Workbook* Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material

**Chemistry: 1,001 Practice Problems For Dummies (+ Free Online Practice)** Heather Hattori, Richard H. Langley, 2014-03-11 Practice makes perfect and helps deepen your understanding of chemistry Every high school requires a course in chemistry and many universities require the course for majors in medicine engineering biology and various other sciences 1001 Chemistry Practice Problems For Dummies provides students of this popular course the chance to practice what they learn in class deepening their understanding of the material and allowing for supplemental explanation of difficult topics 1001 Chemistry Practice Problems For Dummies takes you beyond the instruction and guidance offered in Chemistry For Dummies giving you 1 001 opportunities to practice solving problems from the major topics in chemistry Plus an online component provides you with a collection of chemistry problems presented in multiple choice format to further help you test your skills as you go Gives you a chance to practice and reinforce the skills you learn in chemistry class Helps you refine your understanding of chemistry Practice problems with answer explanations that detail every step of every problem Whether you re studying chemistry at the high school college or graduate level the practice problems in 1001 Chemistry Practice Problems For Dummies range in areas of difficulty and style providing you with the practice help you need to score high at exam time

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