

Thermochemistry Problems - Worksheet Number Two (answers available on web site)

The older energy unit of calories has not been discussed in class. You may see it from time to time. The conversion is $1.000 \text{ cal} = 4.184 \text{ J}$. All the calculation techniques are the same regardless of using calories or Joules.

- Convert from one unit to the other:
 - 1.69 Joules to calories
 - 0.3587 J to cal
 - 820.1 J to kilocalories
 - 68 calories to kilocalories
 - 423 calories to kilocalories
 - 20.0 calories to Joules
 - 252 cal to J
 - 2.45 kilocalories to calories
 - 556 kilocalories to cal
 - 6.78 kilocalories to kilojoules
 - 59.6 calories to kcal
 - 449.6 joules to kilojoules
 - 9.806 kJ to J
 - 5.567 cal to J
 - 5467.9 kcal to J
- Determine the temperature change when:
 - 20.0 g of water is heated from 16.8°C to 39.2°C .
 - 35.0 g of water is cooled from 56.5°C to 5.9°C .
 - 50.0 g of liquid water is heated from 0.0°C to 100.0°C .
 - 25.0 g of ice is warmed from -25.0°C to 0.0°C , but does not melt.
 - 30.0 g of steam heats from 373.2 K to 405.0 K .
- Determine the energy required (in Joules) when the temperature of 3.21 grams of water increases by 4.0°C while remaining liquid.
- Determine the energy needed (in Joules) when 55.6 grams of water at 43.2°C is heated to 78.1°C .
- Determine the energy required (in kilojoules) when cooling 456.2 grams of water at 89.2°C to a final temperature of 5.9°C .
- Determine the energy required to:
 - melt 5.62 moles of ice at 0°C .
 - melt 74.5 grams of ice at 0°C .
 - boil 0.345 moles of water at 100.0°C .
 - boil 43.89 grams of water at 100.0°C .
- Determine the energy change involved to:
 - Convert 16.2 grams of ice to liquid water.
 - Convert 5.8 grams of water to steam
 - Convert 98.2 grams of water to ice.
 - Convert 52.6 grams of steam to water
 - Convert 34.0 grams of water at 20.0°C to steam at 100.0°C .
 - Convert 125.0 grams of ice at 0.0°C to steam at 100.0°C .
 - Convert 25.9 grams of steam at 100.0°C to ice at 0.0°C .

Thermochemistry Problems Number 2 Answers

Pierre Vogel, Kendall N. Houk



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Chemistry Workbook For Dummies Peter J. Mikulecky, Katherine Brutlag, Michelle Rose Gilman, Brian Peterson, 2008-08-06 From liquids and solids to acids and bases work chemistry equations and use formulas with ease Got a grasp on the chemistry terms and concepts you need to know but get lost halfway through a problem or worse yet not know where to begin Have no fear this hands on guide helps you solve many types of chemistry problems in a focused step by step manner With problem solving shortcuts and lots of practice exercises you ll build your chemistry skills and improve your performance both in and out of the science lab You ll see how to work with numbers atoms and elements make and remake compounds understand changes in terms of energy make sense of organic chemistry and more 100s of Problems Know where to begin and how to solve the most common chemistry problems Step by step answer sets clearly identify where you went wrong or right with a problem Understand the key exceptions to chemistry rules Use chemistry in practical applications with confidence

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Student Solutions Manual to Accompany Atkins' Physical Chemistry Peter Bolgar, Haydn Lloyd, James Keeler, Aimee North, Vladimiras Oleinikovas, Stephanie Smith, 2018 The Student Solutions Manual to accompany Atkins Physical Chemistry 11th Edition provides full worked solutions to the exercises and the odd numbered discussion questions and problems presented in the parent book The manual is intended for students and provides helpful comments and friendly advice to aid understanding

A Complete Preparation for the MCAT Aftab S. Hassan, James L. Flowers, 1992 This guide for MCAT preparation applies the principles of active and problem based learning to an updated review of content and skills with models for enhanced problem solving and critical thinking abilities There are details on setting up a self managed study programme with guidelines for time management and stress management All areas tested on the exam are covered verbal reasoning physical science writing sample biological sciences with practice questions to chart progress

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Problems in Physical Chemistry with Practical Applications Edmund Brydges Rudhall Prideaux, 1912

Chemical Thermodynamics J. Rouquerol, R. Sabbah, 2013-10-22 Chemical Thermodynamics 4 presents the

application of experimental methods of chemical thermodynamics This book discusses the three properties of biological molecules namely colossal dimension exclusive orderliness and capability to be in different states or conformations depending on conditions Organized into eight chapters this book begins with an overview of the trends in thermochemistry that involve complex reaction systems and product mixtures This text then discusses the problems relating to the standard state of solids and illustrates the utilization of enthalpy of mixing data Other chapters consider the available heat capacity results in the liquid gas This book discusses as well the high temperature measurement of thermodynamic data for substances of metallurgical interest The final chapter deals with the important advances in the experimental methods of heat capacity measurements including laser flash calorimetry and the high resolution heat capacity calorimeter This book is a valuable resource for chemists physical chemists thermochemists thermophysicists nuclear engineers and research workers

14000 + Objective Questions - General Studies Manohar Pandey, 2018-04-20 Arihant has come up with a revised edition of a compendium of over 14000 questions which will significantly improve the knowledge of aspiring students by providing them with ready and reliable practice material for General Studies The book has been designed for the aspirants preparing for IAS CSAT State PCS CDS NDA and other competitive examinations The revised edition of this question bank focuses on Indian History Culture India World Geography Env Eco Indian Polity Indian Economy General Science Science Technology General Knowledge and Current Affairs The book contains the collection of over 14000 questions covering General Studies The History section covers ancient medieval and modern history whereas the Geography section covers world geography Indian geography and environment ecology The General Science section covers Physics Chemistry Biology and Science Technology The questions covered in the book contain answers side by side to help aspirants evaluate themselves after attempting a certain number of questions Also the questions asked in recent years General Studies examinations have been provided in the book with authentic and detailed solutions to help aspirants get an insight into the recent examination pattern and the types of questions asked therein Each chapter in the book contains a variety of questions according to the latest pattern Assertion Reason Matching Multi Statements Arrangements Pairing etc Also more than 500 questions based on Current Affairs have been provided in the book to give an additional advantage to the aspirants As the book contains ample number of objective questions which have been designed for students of various competitive examinations it for sure will act as the best preparation material for general studies for UPSC CSAT State PCS CDS NDA etc

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