

7. A chemical system that has no output or input of matter but does have an input/output of energy is known as:
- A closed system
  - An open system
  - An isolated system
  - None of the above
8. Which statement concerning the diagram below is true?
- $\Delta H$  is positive
  - The system is endothermic
  - Energy is released to the surroundings
  - The enthalpy of the products is greater than the enthalpy of the reactants
- 
9. In the reaction  $2 \text{NO} (\text{g}) + \text{O}_2 (\text{g}) \rightarrow 2 \text{NO}_2 (\text{g})$ , if the concentration of nitrogen dioxide changes from 0.45 mol/L to 1.00 mol/L in 2 minutes, what is the overall rate of production of nitrogen dioxide in the system?
- 3.64 mol/(L•min)
  - 0.275 mol/(L•min)
  - 0.137 mol/(L•min)
  - 12.6 mol/(L•min)
  - 0.333 mol/(L•min)
10. The activated complex...
- is an unstable molecule
  - has the maximum potential energy possible
  - may revert to reactants
  - all of the above
11. A catalyst works by:
- changing the products that are formed in the reaction
  - decreasing the activation energy of the reaction
  - decreasing the enthalpy change of the reaction
  - changing the enthalpy of the reactants
  - increasing the activation energy of the reaction
12. Which of the following could be a rate law for a fourth order reaction?
- $r = k[\text{X}]^4[\text{Y}]^1$
  - $r = k[\text{X}]^4[\text{Y}]^4$
  - $r = k[\text{X}]^2[\text{Y}]^2$
  - $r = k[\text{X}]^1[\text{Y}]^2$
13. What property could help indicate the reaction rate for the following reaction?
- $$\text{NaOH}_{(\text{s})} \rightarrow \text{Na}^+_{(\text{aq})} + \text{OH}^-_{(\text{aq})} \quad \Delta H = -44.51 \text{ kJ/mol}$$
- Change in pressure
  - Change in color
  - Change in temperature
  - Change in mass

# Thermodynamics And Reaction Rates Review Answer

**Jessica J Manson**



## **Thermodynamics And Reaction Rates Review Answer:**

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*Reviews in Computational Chemistry, Volume 4* Kenny B. Lipkowitz, Donald B. Boyd, 2009-09-22 This volume in the series brings together renowned experts in the field to present the reader with an account of the latest developments in quantum mechanics molecular dynamics and the teaching of computational chemistry There are so many developments in the field of computational chemistry that it is difficult to keep track of them The series was established to review the high volume of developments in the field Rather than create a traditional article each author approaches a topic to enable the reader to understand and solve problems and locate key references quickly Each article has tutorial value An updated compendium of software for molecular modeling appears as an appendix as in previous volumes To the editors knowledge this is the most complete listing of sources of software for computational chemistry anywhere

*Geological Sequestration of Carbon Dioxide* Luigi Marini, 2006-10-12 The contents of this monograph are two scope First it intends to provide a synthetic but complete account of the thermodynamic and kinetic foundations on which the reaction path modeling of geological CO<sub>2</sub> sequestration is based In particular a great effort is devoted to review the thermodynamic properties of CO<sub>2</sub> and of the CO<sub>2</sub> H<sub>2</sub>O system and the interactions in the aqueous solution the thermodynamic stability of solid product phases by means of several stability plots and activity plots the volumes of carbonation reactions and especially the kinetics of dissolution precipitation reactions of silicates oxides hydroxides and carbonates Second it intends to show the reader how reaction path modeling of geological CO<sub>2</sub> sequestration is carried out To this purpose the well known high quality EQ3 6 software package is used Setting up of computer simulations and obtained results are described in detail and used EQ3 6 input files are given to guide the reader step by step from the beginning to the end of these exercises Finally some examples of reaction path and reaction transport modeling taken from the available literature are presented The results of these simulations are of fundamental importance to evaluate the amounts of potentially sequestered CO<sub>2</sub> and their evolution with time as well as the time changes of all the other relevant geochemical parameters e g amounts of solid reactants and products composition of the aqueous phase pH redox potential effects on aquifer porosity In other words in this way we are able to predict what occurs when CO<sub>2</sub> is injected into a deep aquifer Provides applications for investigating and predicting geological carbon dioxide sequestration Reviews the geochemical literature in the field Discusses the importance of geochemists in the multidisciplinary study of geological carbon dioxide sequestration

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Standards,1971     *Publications of the National Bureau of Standards* United States. National Bureau of Standards,1983

**Publications of the National Bureau of Standards, 1973 Catalog** United States. National Bureau of Standards,1974

**Chemical Principles** Peter Atkins,Loretta Jones,2007-08 Written for calculus inclusive general chemistry courses Chemical Principles helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry's frontier continually demonstrating how to solve problems think about nature and matter and visualize chemical concepts as working chemists do Flexibility in level is crucial and is largely established through clearly labeling separating in boxes the calculus coverage in the text Instructors have the option of whether to incorporate calculus in the coverage of topics The multimedia integration of Chemical Principles is more deeply established than any other text for this course Through the unique eBook the comprehensive Chemistry Portal Living Graph icons that connect the text to the Web and a complete set of animations students can take full advantage of the wealth of resources available to them to help them learn and gain a deeper understanding     *Publications of the National Institute of Standards and Technology ... Catalog* National Institute of Standards and Technology (U.S.),1974     *Catalog of National Bureau of Standards Publications, 1966-1976: Key word index* United States. National Bureau of Standards. Technical Information and Publications Division,1978     **Nuclear Science Abstracts** ,1973     *Scientific and Technical Aerospace Reports* ,1991 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database     **Catalog of National Bureau of Standards Publications, 1966-1976: Citations and abstracts** United States. National Bureau of Standards. Technical Information and Publications Division,1978

Chemistry of Polysaccharides Gennady Zaikov,2005-04-18 This monograph summarizes scientific achievements in the field of polysaccharide chemistry performed in the last decade For the first time the particular features of polysaccharide structure physical chemical properties and biochemical transformations in their interrelations are considered as well as the questions of polysaccharide modification along the whole hierarchical chain including starch mono di oligo and polysaccharides chitin chitozan and cellulose Polysaccharide interactions with diverse solvents and low molecular macrocyclic and high molecular substances in solutions are scrutinized Several aspects of mechanical chemical modification of polysaccharides in solid state and in gel are observed Modern concepts of enzyme degradation of polysaccharides and technologies of natural polymers processing are analyzed The book is intended for wide range of readers students post graduates engineers and scientists engaged in polymer chemistry organic and physical chemistry     **Publications of the National Bureau of Standards, 1974 Catalog** United States. National Bureau of Standards,1975

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