A 10.0 g piece of copper wire, sitting in the sun reaches a temperature of 80.0°C. How many Joules are released when the copper cools to 40.0°C? The specific heat capacity of copper is 0.377 J/(g.°C).

First: write down your givens

Mass (m) = 10 g

Initial temperature (Ti) = 80°C

Final temperature (Tf) = 40 °C

Specific heat capacity = 0.377 J/(g.°C)

The formula to find heat (Q) (or thermal energy) = $m c \Delta T$

Remember that $\Delta T = Tr - Tr$ Next: Plug your givens in your formula

Q= (10) (0.377) (80-40) Q= 150.8 Joules

 The specific heat capacity of water is 4.184 J/(g. °C). How much thermal energy is required to change the temperature of 700.0g of water from 25.6 °C to 75.4 °C?

Givens: c = 4.184 J/(g."C) m = 700 g $T_f = 75.4\text{"C}$ $T_i = 25.6\text{"C}$ The formula for thermal energy is $Q = m c \Delta T$ Q = (700) (4.184) (75.4-25.6) Q = 145, 854.24 Joules

Thermal Energy Problems Answer Key

Arshad Iqbal

Thermal Energy Problems Answer Key:

Ultra-High Temperature Thermal Energy Storage, Transfer and Conversion Alejandro Datas, 2020-09-01 Ultra High Temperature Thermal Energy Storage Transfer and Conversion presents a comprehensive analysis of thermal energy storage systems operating at beyond 800 C Editor Dr Alejandro Datas and his team of expert contributors from a variety of regions summarize the main technological options and the most relevant materials and characterization considerations to enable the reader to make the most effective and efficient decisions. This book helps the reader to solve the very specific challenges associated with working within an ultra high temperature energy storage setting It condenses and summarizes the latest knowledge covering fundamentals device design materials selection and applications as well as thermodynamic cycles and solid state devices for ultra high temperature energy conversion This book provides a comprehensive and multidisciplinary guide to engineers and researchers in a variety of fields including energy conversion storage cogeneration thermodynamics numerical methods CSP and materials engineering It firstly provides a review of fundamental concepts before exploring numerical methods for fluid dynamics and phase change materials before presenting more complex elements such as heat transfer fluids thermal insulation thermodynamic cycles and a variety of energy conversation methods including thermophotovoltaic thermionic and combined heat and power Reviews the main technologies enabling ultra high temperature energy storage and conversion including both thermodynamic cycles and solid state devices Includes the applications for ultra high temperature energy storage systems both in terrestrial and space environments Analyzes the thermophysical properties and relevant experimental and theoretical methods for the analysis of high temperature materials

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