

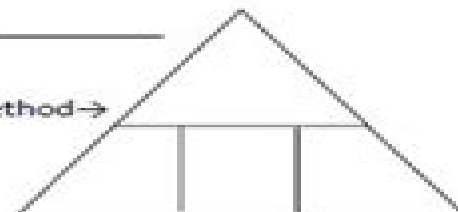
# Thermochemistry Unit Review Packet

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Part 1 Specific Heat Equation Problems

Set up and use the triangle method →



The specific heat equation is  $Q = m \cdot c \cdot \Delta T$ .

$Q$  = heat energy,  $m$  = mass, and  $\Delta T$  = change in temperature ( $T_{\text{final}} - T_{\text{initial}}$ )

Show your work and label all of the units of measurement throughout.

1. How many joules of heat are needed to raise the temperature of 52.0 g of zirconium from 25°C to 56°C, if the specific heat of zirconium is 0.27 J/g°C?
2. A 3.65 g piece of iron absorbs 144.7 joules of heat energy and its temperature changes from 23.0°C to 111°C. Calculate the specific heat capacity of iron.

## Part 2 Gibbs Free Energy Equation Problems

$$\Delta G = \Delta H - T\Delta S$$

$$\Delta H = T\Delta S$$

$$\Delta S = \frac{\Delta H}{T}$$

$$^{\circ}\text{C} + 273.2 = ^{\circ}\text{K}$$

1. For the reaction  $\text{NH}_4\text{Br}(s) \rightarrow \text{NH}_3(g) + \text{HBr}(g)$ , at 302.7°K, the enthalpy change is 158 kJ/mol, and the change in entropy is 0.422 kJ/mol·°K. Calculate  $\Delta G$ .
2. Calculate the entropy of the vaporization phase change for silver at 2453°K. The change in enthalpy is +250.58 kJ/mol.

# Thermochemistry Review Packet

**Martin Quack, Frederic Merkt**



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**Authorization of the Standard Reference Data Act and Review of the National Bureau of Standards** United States. Congress. Senate. Committee on Commerce, Science, and Transportation. Subcommittee on Science, Technology, and Space, 1978      The Big Book of Chemistry Teacher Stories Jeff Lark, Stories from years of teaching high school chemistry

*NBS Special Publication*, 1968      **Publications of the National Bureau of Standards** United States. National Bureau of Standards, 1983      Publications of the National Bureau of Standards ... Catalog United States. National Bureau of

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Standards. Technical Information and Publications Division, 1978      **Scientific and Technical Aerospace Reports**, 1994

*Meteorites and the Early Solar System II* Dante S. Lauretta, Harry Y. McSween, 2006-07-01 They range in size from microscopic particles to masses of many tons The geologic diversity of asteroids and other rocky bodies of the solar system are displayed in the enormous variety of textures and mineralogies observed in meteorites The composition chemistry and mineralogy of primitive meteorites collectively provide evidence for a wide variety of chemical and physical processes This book synthesizes our current understanding of the early solar system summarizing information about processes that occurred before its formation It will be valuable as a textbook for graduate education in planetary science and as a reference for meteoriticists and researchers in allied fields worldwide      **Technical Abstract Bulletin**, 1979      Catalog of National Bureau of Standards Publications, 1966-1976: pt. 1-2. Key word index United States. National Bureau of Standards, 1978

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