

COLLABORATIVE ACTIVITY: Think-Pair-Share

PROBLEM SOLVING:

1) The combustion of a sample of methane, CH_4 , (g) at 1 bar, produced 2.46 grams of water:



a) How many grams of methane are used?

b) How much oxygen was used up in grams?

2) Using the following equation:



How many grams of sodium sulfate will be formed if you start with 200.0 grams of sodium hydroxide and you have an excess of sulfuric acid?

3) Using the following equation:



How many grams of lithium sulfate will be needed to make 250.0 grams of lithium sulfate, assuming that you have an adequate amount of lead (II) sulfate to do the reaction?

4) Using the following equation:



Calculate how many grams of iron can be made from 16.5 grams of Fe_2O_3 by the following equation.

5) Using the following equation:



Calculate how many grams of lithium are needed to prepare 28.6 grams of Nitrogen gas.

6) Using the following equation:



How many grams of KMnO_4 are needed to carry out this reaction on 11.4 grams of KNO_3 ?

7) Using the following equation:



How many grams of oxygen (O_2) are needed to react with 56.8 grams of ammonia by this reaction?

8) Using the following equation:



Calculate the amount of grams of water (H_2O) that can be made this way from 16.4 grams of Na_2O_2 .

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Michael Brown



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