

HEATING CURVES QUESTIONS

PART A – HEATING CURVES. Use the heating curve to the right to answer the following questions:

1. What is the melting point of the substance? **B/ 5 C**
2. What is the boiling point of the substance? **D/ 15 C**
3. Which letter represents heating of the solid? **A**
4. Which letter represents heating of the gas? **E**
5. Which letter represents melting of the solid? **B**
6. Which letter represents boiling of the liquid? **D**
7. At points b and d why doesn't the temperature change even though heat energy is added?

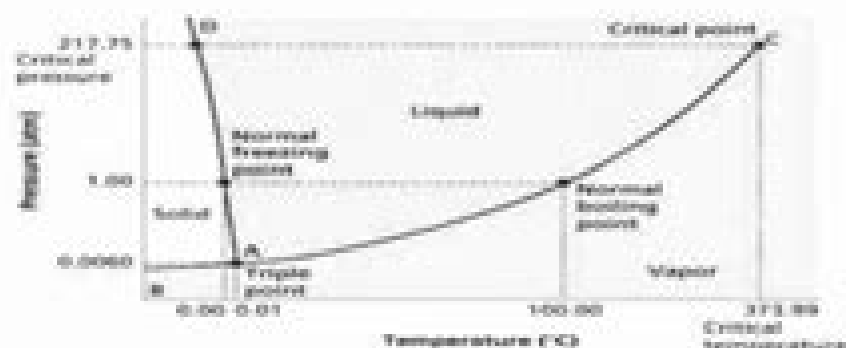


The Substance must absorb enough heat/ kinetic energy to Escape the intermolecular actions/ bonds (i.e. hydrogen bonds) that Keep them their current phase/ state

Refer to the phase diagram below when answering the questions on this worksheet

PART C – PHASE DIAGRAMS. Use the phase diagram for water below to answer the following questions.

8. What is the state of water at 2 atm and 50°C?
LIQUID
9. What phase change will occur if the temperature is lowered from 80°C to -5°C at 1 atm?
FREEZING
10. You have ice at -10°C and 1 atm. What could you do in order cause the ice to sublime?
Decrease the pressure to less than 0.060 atm



Predict the physical state of each of these substances at the indicated temperature. Use the melting point and boiling point data from the table below.

- a. phenol at 99°C
- b. ammonia at -25°C
- c. methanol in an ice-water bath
- d. methanol in a boiling-water bath
- e. ammonia at -100°C
- f. phenol at 25°C

- a. **Liquid**
- b. **Gas**
- c. **Liquid**
- d. **Gas**
- e. **Solid**
- f. **Solid**

Review Activity A Heating Curve Answers

Minjie Lin

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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