

SECTION 2-1

REVIEW AND REINFORCE

Solids, Liquids, and Gases

◆ Understanding Main Ideas

Use the diagram to answer Questions 1 through 3.
Write your answers on a separate sheet of paper.

1. Identify the physical state of the substances pictured.
2. What would happen to the shape of each substance if the jars were broken? Use the differences in the physical state of the substances to explain your answer.
3. Would the volume of each substance change if each were moved into a larger container? Explain.



2

◆ Building Vocabulary

Write a definition for each of the following terms in the spaces provided.

4. solid particles are packed very close together & stay in a fixed position, definite shape & volume.
5. liquid particles are slightly less packed than solids, they can freely move past each other, definite volume, indefinite shape.
6. gas particles spread apart filling all space available to them, moving at high speeds in all directions, indefinite shape, indefinite volume.
7. viscosity resistance of a liquid to flowing. the higher the viscosity, the slower the flow.
8. amorphous solid particles are not arranged in a regular pattern, no definite melting point.
9. crystalline solid particles are in a regular repeating pattern forming crystals, definite melting point.
10. fluid a substance that flows, a liquid or gas.

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RJ Shavelson



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