



Name: _____

Date: _____

Student Exploration: Ionic Bonds

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Nate gives one marker to Clara.

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Maggie gives one to Fred and then Florence.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *Ionic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** (▶) to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain electrons. The outermost electrons are called **valence electrons**. (Inner electrons are not shown.)

How many valence electrons does each atom have? Sodium: **1** Chlorine: **7**

2. Click **Pause** (⏸). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

- A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal? **Sodium** Which is a nonmetal?

Chlorine



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S Ben Porath



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Prentice Hall Science Explorer Michael J. Padilla,2002

In Situ Molecular Pathology and Co-expression Analyses

Gerard J. Nuovo,2020-09-03 Major advancements in the field of in situ molecular pathology have occurred since publication of the first edition In Situ Molecular Pathology and Co expression Analyses Second Edition continues to teach both the molecular basis for the improvements and the actual protocols This is the unique feature that separates it from the pack of other cook book type approaches The fields of in situ hybridization and immunohistochemistry have expanded rapidly where computer based analyses systems have greatly expanded the power of these methods Further knowledge of the marked improvements in the reagents themselves since the first edition can make the difference of excellent versus misleading data The automated platforms require that researchers and diagnostic biomedical investigators have a good understanding of the basics of in situ based tests protocols and biochemistry for troubleshooting in order to maximize the use of these platforms This second edition focuses attention on straightforward protocols used to simultaneously detect two or more proteins nucleic acids within intact tissue by doing co expression analyses Practicing molecular pathologists diagnostic pathologists laboratory directors and toxicologists as well as clinicians and researchers in training will benefit from this clear presentation of protocols and theoretical framework Data derived from in situ hybridization and immunohistochemistry Explains the theory and foundation of immunohistochemistry and in situ hybridization and presents easy to follow experimental protocols with tricks of the trade Includes two new chapters Recent improvements in immunohistochemistry and in situ hybridization Quality control for immunohistochemistry and in situ hybridization How to know if the color change is signal or background The second edition also includes a detailed test to help one learn the basics of histologic interpretation of tissues and a separate detailed test in how to differentiate signal from background Includes chapter ending summaries of Key Points to Remember bringing beginners up to speed with any seasoned veteran in the field Thoughtfully tackles the molecular basis if IHC and ISH along with application of that knowledge to improving the techniques is significant **The Science Teacher** ,1996 Some issues are accompanied by a CD ROM on a selected topic **Biology** Gerhart Campbell,2000-09

Visualizing Everyday Chemistry Douglas P. Heller,Carl H. Snyder,2015-01-20 Visualizing Everyday Chemistry Binder Ready Version is for a one semester course dedicated to introducing chemistry to non science students It shows what chemistry is and what it does by integrating words with powerful and compelling visuals and learning aids With this approach students not only learn the basic principles of chemistry but see how chemistry impacts their lives and society The goal of Visualizing Everyday Chemistry Binder Ready Version is to show students that chemistry is important and relevant not because we say it is but because they see it is This text is an unbound binder ready version **The Software**

Encyclopedia 2000 Bowker Editorial Staff,2000-05

Bulletin of the Atomic Scientists ,1970-06

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