

## Chemical Bonding

### SECTION 3

**SHORT ANSWER** Answer the following questions in the space provided.

- A** The notation for sodium chloride, NaCl, stands for one  
 (a) formula unit. (c) crystal.  
 (b) molecule. (d) atom.
- D** In a crystal of an ionic compound, each cation is surrounded by a number of  
 (a) molecules. (c) dipoles.  
 (b) positive ions. (d) negative ions.
- B** Compared with the neutral atoms involved in the formation of an ionic compound, the crystal lattice that results is  
 (a) higher in potential energy. (c) equal in potential energy.  
 (b) lower in potential energy. (d) unstable.
- B** The lattice energy of compound A is greater in magnitude than that of compound B. What can be concluded from this fact?  
 (a) Compound A is not an ionic compound.  
 (b) It will be more difficult to break the bonds in compound A than those in compound B.  
 (c) Compound B has larger crystals than compound A.  
 (d) Compound A has larger crystals than compound B.
- B** The forces of attraction between molecules in a molecular compound are generally  
 (a) stronger than the attractive forces among formula units in ionic bonding.  
 (b) weaker than the attractive forces among formula units in ionic bonding.  
 (c) approximately equal to the attractive forces among formula units in ionic bonding.  
 (d) equal to zero.
- Describe the force that holds two ions together in an ionic bond.

**Very strong overall attraction between positive and negative charges**

- What type of energy best represents the strength of an ionic bond?

**Lattice energy**

- What types of bonds are present in an ionic compound that contains a polyatomic ion?

# Review Chemical Bonding Section 5 Answer Key

**The Princeton Review**



## **Review Chemical Bonding Section 5 Answer Key:**

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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Review Chemical Bonding Section 5 Answer Key**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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