

## Unit 8 Review

Write *increases* or *decreases* to complete each relationship.

1. As volume increases, pressure \_\_\_\_\_ decreases \_\_\_\_\_.
2. As volume increases, temperature \_\_\_\_\_ increases \_\_\_\_\_.
3. As pressure increases, temperature \_\_\_\_\_ increases \_\_\_\_\_.
4. As pressure increases, concentration \_\_\_\_\_ increases \_\_\_\_\_.
5. As force decreases, pressure \_\_\_\_\_ decreases \_\_\_\_\_.

True or False.

6. \_\_\_\_ T \_\_\_\_ The Kinetic Theory of Matter deals with moving particles.
7. \_\_\_\_ T \_\_\_\_ Gases can be collected by trapping them over water.
8. \_\_\_\_ T \_\_\_\_ Real gases exist in the world around us.
9. \_\_\_\_ T \_\_\_\_ Densities of gases are lower than 1 g/mL.
10. \_\_\_\_ F \_\_\_\_ If the barometric pressure is falling, that means good weather is coming.
11. \_\_\_\_ T \_\_\_\_ The troposphere contains the most air particles.
12. \_\_\_\_ T \_\_\_\_ Temperature is a measure of how fast particles are moving.

13. The gas that makes up the highest percentage of the atmosphere is

\_\_\_\_ nitrogen \_\_\_\_\_.

14. Atmospheric pressure is measured using a \_\_\_\_ barometer \_\_\_\_\_.

15. Squeezing gas particles together is known as \_\_\_\_ compressibility \_\_\_\_\_.

16. When particles collide with the sides of a container, they create \_\_\_\_ pressure \_\_\_\_\_.

17. The spreading of a gas through a small opening is \_\_\_\_ effusion \_\_\_\_\_.

18. The number of gas particles present is called the \_\_\_\_ concentration \_\_\_\_\_.

Gas Law Problems. Write the answer with the correct unit.

19. A gas has a beginning pressure of 2.95 atm and a volume of 5.15 L. If the ending pressure is 4.23 atm, what is the ending volume?  $P_1V_1 = P_2V_2$   
 $(2.95)(5.15) \div 4.23 = 3.59$

20. A gas has a beginning temperature of 300 K and a volume of 4.06 L. If the ending temperature is 176 K, what is the ending volume?  $V_1T_2 = V_2T_1$   
 $(4.06)(176) \div 300 = 2.38$

21. A gas has a beginning pressure of 2.09 atm, a temperature of 268 K, and a volume of 0.450 L. If the ending pressure is 1.2 atm and the temperature is 149 K, what is the ending volume?  $P_1V_1T_2 = P_2V_2T_1$

# Review Section 1 Gases Answers

**Cynthia Y. Young**



## **Review Section 1 Gases Answers:**

**Justice of the Peace and Local Government Review**, 1892      **Biological Weapons** Kristy Young Johnson, Paul Matthew Nolan, 2016-05-03 Gives readers a detailed understanding of how specific biological weapons work and how those affected by the weapons would be treated Teaches the reader to recognize the symptoms of each biological weapon and understand the threat these weapons pose Concentrates on the weapons considered the greatest threats by the CDC such as Anthrax Botulism Smallpox Ricin toxin Ebola Plague and Viral encephalitis Provides a detailed understanding of how specific biological weapons work and how to recognize the symptoms of those affected by the weapons as well as how they would be treated Includes case studies chapter review questions and the instructor s supplemental materials include PowerPoint presentations a Test Bank and suggestions for student projects Begins with a primer on microbiology the human immune system s response to these biological agents and the defense agencies involved with protecting the public against these agents      *Selected Topics in Superconductivity* L. C. Gupta, M. S. Multani, 1993 *Selected Topics in Superconductivity* reflects the high level of activity in the discovery of high T<sub>c</sub> superconductivity Out of the 19 articles that it has a fairly good number of them discuss several important and fundamental aspects of the high T<sub>c</sub> superconductivity Some of the issues related to the phenomenon of superconductivity in general are discussed in a few of the manuscripts The remaining articles deal with superconductivity in unconventional and highly correlated metals      **Applied Mechanics Reviews**, 1974      Fire and Emergency Services Instructor: Principles and Practice Forest F Reeder, Alan E Joos, 2019-03-28 The National Fire Protection Association NFPA the International Association of Fire Chiefs IAFC and the International Society of Fire Service Instructors ISFSI are pleased to bring you *Fire and Emergency Services Instructor Principles and Practice* Third Edition With a full library of technological resources to engage candidates and assist instructors *Fire and Emergency Services Instructor* takes training off the printed page This text meets and exceeds all of the job performance requirements JPRs for Fire and Emergency Services Instructor I II and III as well as two new levels for Live Fire Instructor and Live Fire Instructor in Charge of the 2019 Edition of NFPA 1041 Standard for Fire and Emergency Services Instructor Professional Qualifications Innovative features include Rapid access of content through clear and concise Knowledge and Skills Objectives with page number references and NFPA 1041 correlations Promotion of critical thinking and classroom discussion through the Training Bulletin and Incident Report features JPRs in Action feature identifying the specific responsibilities of the Fire and Emergency Services Instructor I II and III relating to the job performance requirements JPRs Tips geared toward the company level instructor department training officer and training program manager offering instruction techniques test writing and evaluation pointers and helpful notes on communication and curriculum delivery Realistic instructor scenarios with questions designed to provoke critical thinking in the learning environment New to the Third Edition In depth discussion of student centered learning Learner centered teaching methods and strategies Evidence based techniques for improving

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Gas Dynamics maintains the focus on gas flows below hypersonic. This targeted approach provides a cohesive and rigorous examination of most practical engineering problems in this gas dynamics flow regime. The conventional one dimensional flow approach together with the role of temperature entropy diagrams are highlighted throughout. The authors noted experts in the field include a modern computational aid, illustrative charts and tables and myriad examples of varying degrees of difficulty to aid in the understanding of the material presented. The updated edition of Fundamentals of Gas Dynamics includes new sections on the shock tube, the aerospoke nozzle and the gas dynamic laser. The book contains all equations, tables and charts necessary to work the problems and exercises in each chapter. This book's accessible but rigorous style offers a comprehensively updated edition that includes new problems and examples. Covers fundamentals of gas flows targeting those below hypersonic. Presents the one dimensional flow approach and highlights the role of temperature entropy diagrams. Contains new sections that examine the shock tube, the aerospoke nozzle, the gas dynamic laser and an expanded coverage of rocket propulsion. Explores applications of gas dynamics to aircraft and rocket engines. Includes behavioral objectives, summaries and check tests to aid with learning. Written for students in mechanical and aerospace engineering and professionals and researchers in the field, the third edition of Fundamentals of Gas Dynamics has been updated to include recent developments in the field and retains all its learning aids. The calculator for gas dynamics calculations is available at <https://www.oscarbiblarz.com/gascalculator> gas dynamics calculations. *Toxicological Profile for Mustard Gas*, 1992

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