

## Unit 8 Review

Write *increases* or *decreases* to complete each relationship.

1. As volume increases, pressure \_\_\_\_\_decreases\_\_\_\_\_.
2. As volume increases, temperature \_\_\_\_\_increases\_\_\_\_\_.
3. As pressure increases, temperature \_\_\_\_\_increases\_\_\_\_\_.
4. As pressure increases, concentration \_\_\_\_\_increases\_\_\_\_\_.
5. As force decreases, pressure \_\_\_\_\_decreases\_\_\_\_\_.

True or False.

6. \_\_\_\_T\_\_\_\_\_ The Kinetic Theory of Matter deals with moving particles.
7. \_\_\_\_T\_\_\_\_\_ Gases can be collected by trapping them over water.
8. \_\_\_\_T\_\_\_\_\_ Real gases exist in the world around us.
9. \_\_\_\_T\_\_\_\_\_ Densities of gases are lower than 1 g/mL.
10. \_\_\_\_F\_\_\_\_\_ If the barometric pressure is falling, that means good weather is coming.
11. \_\_\_\_T\_\_\_\_\_ The troposphere contains the most air particles.
12. \_\_\_\_T\_\_\_\_\_ Temperature is a measure of how fast particles are moving.

13. The gas that makes up the highest percentage of the atmosphere is

\_\_\_\_nitrogen\_\_\_\_\_.

14. Atmospheric pressure is measured using a \_\_\_\_barometer\_\_\_\_\_.

15. Squeezing gas particles together is known as \_\_\_\_compressibility\_\_\_\_\_.

16. When particles collide with the sides of a container, they create \_\_\_\_pressure\_\_\_\_\_.

17. The spreading of a gas through a small opening is \_\_\_\_effusion\_\_\_\_\_.

18. The number of gas particles present is called the \_\_\_\_concentration\_\_\_\_\_.

Gas Law Problems. Write the answer with the correct unit.

19. A gas has a beginning pressure of 2.95 atm and a volume of 5.15 L. If the ending pressure is 4.23 atm, what is the ending volume?  $P_1V_1 = P_2V_2$   
 $(2.95)(5.15) \div 4.23 = 3.59$

20. A gas has a beginning temperature of 300 K and a volume of 4.06 L. If the ending temperature is 176 K, what is the ending volume?  $V_1T_2 = V_2T_1$   
 $(4.06)(176) \div 300 = 2.38$

21. A gas has a beginning pressure of 2.09 atm, a temperature of 268 K, and a volume of 0.450 L. If the ending pressure is 1.2 atm and the temperature is 149 K, what is the ending volume?  $P_1V_1T_2 = P_2V_2T_1$

# Review Section 1 Gases Answers

**RJ Shavelson**



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