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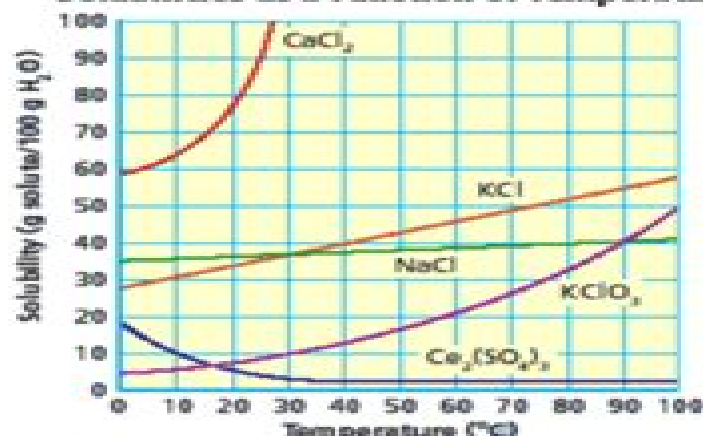
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Solubility Curve Worksheet 2



Directions: Use the graph below to answer the following questions.

Solubilities as a Function of Temperature



Types of Solutions

- > Points that collectively make up the solubility curve (points **ON** the curve) represent _____ solutions.
- > Points that are **BELOW** the curve represent _____ solutions.
- > Points **ABOVE** the solubility curve represent _____ solutions, and the **difference** between the point above the curve and on the curve represents the amount of solute which will precipitate out.
- > Say if the resulting solutions would be saturated, supersaturated or unsaturated.
 - o 60 g of KCl at 70°C _____
 - o 10 g of KClO₃ at 60°C _____
 - o 70 g of CaCl₂ at 20°C _____

1. a. What is the solubility of calcium chloride (CaCl₂) at 5°C? _____
 b. What is the solubility of calcium chloride at 25°C? _____
 c. What is happening to the solubility and temperature for CaCl₂? Solubility is _____ as temperature is _____.
2. a. What is the solubility of cerium sulfate - Ce₂(SO₄)₃ at 10°C? _____
 b. What is the solubility of cerium sulfate at 50°C? _____
 c. What is happening to the solubility and temperature for Ce₂(SO₄)₃? Solubility is _____ as temperature is _____.
3. Which substance on the graph is **least** soluble at 10°C? _____
4. Which substance on the graph shows the **least** change in solubility from 0°C to 100°C? _____
5. Which solution is more concentrated:
 - a) At 10°C, a saturated solution of NaCl or a saturated solution of CaCl₂ (circle one)
 - b) At 55°C, a saturated solution of KClO₃ or a saturated solution of Ce₂(SO₄)₃ (circle one)
6. What is the mass of cerium sulfate that will dissolve in 50 g of water at 10 degrees Celsius?
7. What is the mass of potassium chloride that will dissolve in 250 g of water at 70 degrees Celsius?
8. At 90 degrees Celsius, 10 g of potassium chlorate is dissolved in 100 g of water. Is this solution saturated, unsaturated, or supersaturated? How do you know?
9. A saturated solution of potassium chlorate is dissolved in 100g of water. If the saturated solution is cooled from 90 degrees Celsius to 60 degrees Celsius, how many grams would crystallize out?

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A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the Princeton Review logo.

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Table of Contents Using Solubility Curves Answers

1. Understanding the eBook Using Solubility Curves Answers
 - The Rise of Digital Reading Using Solubility Curves Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Using Solubility Curves Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Using Solubility Curves Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Using Solubility Curves Answers
 - Personalized Recommendations
 - Using Solubility Curves Answers User Reviews and Ratings
 - Using Solubility Curves Answers and Bestseller Lists
5. Accessing Using Solubility Curves Answers Free and Paid eBooks
 - Using Solubility Curves Answers Public Domain eBooks
 - Using Solubility Curves Answers eBook Subscription Services
 - Using Solubility Curves Answers Budget-Friendly Options
6. Navigating Using Solubility Curves Answers eBook Formats

- ePub, PDF, MOBI, and More
- Using Solubility Curves Answers Compatibility with Devices
- Using Solubility Curves Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Using Solubility Curves Answers
 - Highlighting and Note-Taking Using Solubility Curves Answers
 - Interactive Elements Using Solubility Curves Answers
- 8. Staying Engaged with Using Solubility Curves Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Using Solubility Curves Answers
- 9. Balancing eBooks and Physical Books Using Solubility Curves Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Using Solubility Curves Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Using Solubility Curves Answers
 - Setting Reading Goals Using Solubility Curves Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Using Solubility Curves Answers
 - Fact-Checking eBook Content of Using Solubility Curves Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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