



SAMPLE QUESTION PAPER - 3 (POST MID TERM 2024)

Class 12 - Chemistry

Time Allowed: 3 hours

Maximum Marks: 70

Section A

1. How much electricity is required in coulomb for the oxidation of 1 mol of FeO to Fe₂O₃? [1]
a) 95000C
b) 96000C
c) 96500C
d) 95550C
2. Saturated solution of KNO₃ is used to make 'salt bridge' because [1]
a) KNO₃ is highly soluble in water
b) velocity of K⁺ is greater than that of NO₃⁻
c) velocity of NO₃⁻ is greater than that of K⁺
d) velocity of both K⁺ and NO₃⁻ are nearly the same.
3. The rate of the first-order reaction is $0.69 \times 10^{-2} \text{ mol L}^{-1} \text{ min}^{-1}$ and the initial concentration is 0.2 mol L^{-1} the half-life period is: [1]
a) 1200 s
b) 600 s
c) 0.33 s
d) 1 s
4. For the reaction $A \rightarrow \text{products}$, at $[A] = 0.4 \text{ M}$, $t_{1/2} = 24 \text{ s}$ and at $[A] = 0.2 \text{ M}$, $t_{1/2} = 12 \text{ s}$. The unit for the rate constant is [1]
a) s⁻²
b) mol L⁻¹s⁻¹
c) L²/mol²/s
d) s⁻¹
5. The anti pernicious anaemia factor which is a coordination compound of Cobalt is: [1]
a) Cyanocobalamin
b) Haemoglobin
c) Desferrioxime B
d) Carbonic anhydrase
6. [Co(NH₃)₅NO₃]SO₄ and [Co(NH₃)₅SO₄]NO₃ exhibit: [1]
a) ionization isomerism
b) coordination isomerism
c) optical isomerism
d) linkage isomerism
7. When diethyl ether is heated with excess of HI, it produces: [1]
a) ethanol
b) iodoform
c) ethyl iodide
d) methyl iodide
8. One mole of an organic compound 'A' with the formula C₃H₈O reacts completely with two moles of HI to form X and Y. When 'Y' is boiled with aqueous alkali forms Z. Z answers the iodoform test. The compound 'A' is [1]

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