

# Words Their Way Elementary Spelling Inventory Feature Guide

Student Amica

Teacher Miss Jaime Fittall

Grade 7

Date 26-2-09

Words Spelled Correctly: 22 / 25 Feature Points: 58 / 62 Total 80 / 87 Spelling Stage Early Derivational Relations

Stages and gradations Features Words	Emergent		Letter Name			Within Word Patterns			Syllables and Affixes			Derivational Relations		Feature Points	Words Spelled Correctly
	Early	Final	Short Vowels	Diagrams	Blends	Early	Middle	Later	Early	Middle	Later	Early	Middle		
1. bed	b	d	e												
2. ship		p	i	sh											
3. when			e	wh											
4. lump	l		u		mp										
5. float		t			fl	oa									
6. train		n			tr	ai									
7. place					pl	a-e									
8. drive		v			dr	i-e									
9. height					hr	igh									
10. shopping			o	sh				pping							
11. spoil					sp		oi								
12. serving							er	ving							
13. chewed				ch			ew	ed							
14. carries							ar	ies	rr						
15. marched				ch			ar	ed							
16. shower				sh			ow								
17. bottle									ll	le					
18. flavor									v	or					
19. open									p	en					
20. cellar									ll	ar					
21. pleasure											ure	pleas			
22. fortunate							or				ate	fortu	n		
23. confident											ent	confi	d		
24. civilize											ize	civil			

25. opposition											tion	pos			
Totals		7/7	5/5	6/6	7/7	5/5	7/7	5/5	4/5	4/5	4/5	4/5	58/62	22/25	

# Words Their Way Elementary Spelling Inventory Feature Guide

Student Bridget

Teacher Miss Jaime Fittall

Grade 7

Date 26-2-2009

Words Spelled Correctly 22 / 25 Feature Points: 58 / 62 Total 80 / 87 Spelling Stage Late Suffixes and Affixes

Stages and gradations Features Words	Emergent		Letter Name			Within Word Patterns			Syllables and Affixes			Derivational Relations		Feature Points	Words Spelled Correctly		
	Emergent	Early	Short Vowels	Diagrams	Blends	Early	Middle	Late	Early	Middle	Late	Unaccented	Final Syllables			Harder Suffixes	Flower or Roots
		Begin															
1. bed	b	d	e														
2. ship		p	i	sh													
3. when			e	wh													
4. lump	l		u		mp												
5. float		t			fl	oa											
6. train		n			tr	ai											
7. place					pl	a-e											
8. drive		v			dr	i-e											
9. bright					br	igh											
10. shopping			a	sh					pping								
11. spoil					sp		oi										
12. serving							er	ving									
13. chewed				ch			ew	ed									
14. carries							ar	ies	rr								

# Sample Graded Feature Guide Words Their Way

**Jennifer Katz**



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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the iron metal dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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75 and higher the coordination of trimethylamine by copper ii

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate  $\text{CuSO}_4$  and 2 0153 grams of iron powder  $\text{Fe}$  the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web stoichiometric principles can be used to determine the chemical equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction the iron ions dissolve in the copper ii sulfate solution to produce copper ii ions

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web under the finalization of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away copper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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web stoichiometry lab iron and copper sulfate sample data data and observations mass of copper ii sulfate pentahydrate 3 27 g of g mass of filter paper mass of filter paper and dry copper mass of dry copper 1 28 g

