

PART-A

Language Proficiency

(Expected Answering Time: 20 Minutes)

Direction: (Question No: 1-5)

Read the following passage and answer the questions that follow.

If students feel excited about exploring data and enjoy learning Statistics and Mathematics, then they should pursue a career in Actuarial Science. This branch of science is all about measuring, managing and modelling financial risk and uncertainties. It is a multidisciplinary field combining Mathematics, Statistics, Finance and Economics. Actuaries use their expertise in three areas to assess and manage the financial impact of uncertain events such as natural disasters, accidents and illnesses. They use various tools and techniques to perform their work such as probability theory, statistics, computer modelling, specialised software and databases. The future career prospects for actuaries are very positive as they have openings in banking, energy sector, telecommunication, insurance companies and consulting firms. There are strong international career prospects for them. Actuaries are in high demand because they possess a unique set of skills that are highly valued by employers across the world. But for becoming an Actuary in India one needs to clear a series of professional examinations offered by the Institute of Actuaries of India (IAI) which is the professional body that governs the actuarial profession in India. It is worth noting that the process of becoming an Actuary in India is quite challenging, and it requires a significant investment of time and effort.

1. Choose the best reason for great demand of the Actuaries.
 - (1) They have mastery in the multidisciplinary fields.
 - (2) They have expertise in using various tools and techniques.
 - (3) They are good in computer operation.
 - (4) They possess a unique set of skills useful in employment sector.
2. Interest in which of the following disciplines is a primary requisite for selecting Actuarial science for academic pursuit?
 - (1) Physical Science
 - (2) Mathematics
 - (3) Mathematics and Statistics
 - (4) Finance and Economics
3. Which of the following is most useful for assessing the financial impact for the event of natural disaster?
 - (1) Computer Modelling
 - (2) Probability Theory
 - (3) Specialised software designing
 - (4) Statistical expertise
4. What sort of examination one has to clear in order to become an Actuary?
 - (1) Professional examination
 - (2) Skill based examination
 - (3) Theoretical examination
 - (4) Task based examination
5. Which of the following best explains the meaning of "prospect" used in the passage (The future career prospect)?
 - (1) Forecast
 - (2) Likelihood
 - (3) Expectation
 - (4) Possibility

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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance lithium li has three electrons two fill the 1 s orbital and the third is placed in the 2 s orbital giving an electron configuration of 1 s 2 2 s 1

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the atoms of elements in the same vertical column of the table have similar electron configurations for example consider the alkaline earth elements group iia

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web the electron removed during the ionization of beryllium he 2s 2 2 is an s electron whereas the electron removed during the ionization of boron he 2s 2 2p 1 is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each

period written out these are he 2 helium 1s 2 ne 10 neon 1s 2 2s 2 2p 6

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web q1 what is the electron configuration of the element potassium answer the atomic number of potassium is 19 a neutral potassium element has 19 electrons its electron configuration is 1s 2 2s 2 2p 6 3s 2 3p 6 4s 1 or simply put as 4s 1 q2 what is the total number of electrons that can be filled in s p and d subshells

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web by building up from hydrogen this table can be used to determine the electron configuration for any atom on the periodic table we will now construct the ground state electron configuration and orbital diagram for a selection of atoms in the first and second periods of the periodic table

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web electron configurations are a simple way of writing down the locations of all of the electrons in an atom as we know the positively charged protons in the nucleus of an atom tend to attract negatively charged electrons

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