

Chemistry 101 Final Ivy Tech

1. Diatomic O₂ can react with the element magnesium to form magnesium oxide (MgO). The balanced chemical equation is:



If 4 moles of magnesium totally reacted with more than enough O₂, how many moles of MgO would be expected to form?

- A. 8 Moles B. 2 Moles C. 4 Moles D. 1 Mole
E. Not Enough Info: C

2. Which of the following is TRUE?

A. Stoichiometry allows prediction of how much of the reactants are necessary to form a given amount of product.

B. All of the above are true.

C. Stoichiometry allows prediction of the amount of products that form in a chemical reaction based on the amounts of reactants.

D. Stoichiometry shows the numerical relationship between chemical quantities in a balanced chemical equation.: B

Sample Questions On Chemistry 101

Marco Cascella



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