# Electron Configuration

### Learning Objectives:

State that electrons are found in energy levels of an atom and I can state the maximum number of electrons in the first three energy levels.

8B Write and determine the standard electronic configuration notation from a diagram for the first 20 elements.

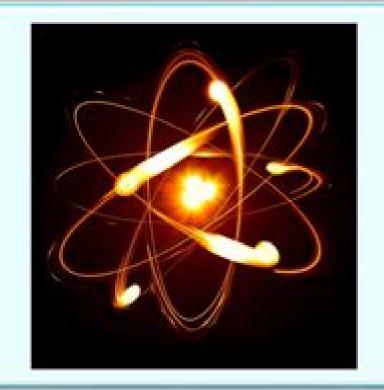
8C Explain why elements in the same group react in a similar way.

### Starter: In your books

Describe the atom to a year 7 who knows nothing about chemistry

Why are atoms 'electronically neutral'?

Why is the periodic table useful for beginner scientists and professionals?



TIF: Make predictions for how an element will react when

### **Teaching Electron Configuration**

**G** Orfield

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