TEACHING TRANSPARENCY WORKSHEET

31

The Activity Series

tice with Chapter 9, Section 9.2

- For each of the following pates of elements, underline the one that would replace the other element in a commonal.
 - st., metalogom, tim-
 - by becoming fluoring
 - c. alterniment, permestors
 - at. wine, positions

- see firms, copper
- ff. icoline, chilocine
- a. Wheel, front
- 2. For each of the following reactions, use the activity series to determine whether the seaction would take place or not. If no reaction takes place, write NN in the black. If a reaction does take place, write the formulas for the products of the reaction. (Hint: N on active metal replaces the hydrogen in water, the hydroxide of the active metal forms.)
 - a. List + FedNO, Jump -- LINDs, + Fe.
 - b. Asin + Hilliag -> No Paris
 - c. CL(g) + KBs(aq) KCL + Bs'a
 - 4. Cabo + AMOND JOSQU Não PORTO
 - w. Agost + HBrosp -- No Party
 - 4. Nico + SeClaber McCla + Trea-
- Use the activity series for eacts to explain why copper metal is used in plumbing where the water might contain compounds of many different ractals.

Con not very appoint - worst report will metallo and

5. The last four metals in the activity series of metals are commonly referred to as the processing metals." Why would these metals be chosen over more active metals for use in codes? Why do you think some near notice metals, such as since or nickel, are sometimes used to color?
Vigital: "Victor, constructing?"

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