

MATH SKILLS TRANSPARENCY WORKSHEET**6****Using the Periodic Table****Use with Chapter 6,
Section 6.2**

1. Identify the number of valence electrons in each of the following elements.

a. Ne _____ e. O _____
b. K _____ f. Cl _____
c. B _____ g. P _____
d. Mg _____ h. Si _____

2. Identify the energy level of the valence electrons in each of the following elements.

a. Br _____
b. N _____
c. Ra _____
d. H _____
e. Ar _____
f. I _____

3. Use the periodic table to write the electron configurations (using noble gas notation) for each of the following elements.

a. Li _____
b. F _____
c. As _____
d. Sr _____
e. Bi _____

4. Determine the group, period, and block of the elements having the following electron configurations.

a. $1s^2$ _____
b. $[\text{Ne}]3s^23p^1$ _____
c. $[\text{Ar}]4s^1$ _____
d. $[\text{Kr}]5s^24d^1$ _____
e. $[\text{Xe}]6s^24f^{14}5d^{10}6p^4$ _____

Teaching Transparency The Periodic Table

Ying-Ying Zheng



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