

## NAMING HYDROCARBONS

Name the compounds below according to the IUPAC naming system

1. $  \begin{array}{ccccccc}  & \text{H} & & \text{H} & & \text{H} & \\  &   & &   & &   & \\  \text{H} & - \text{C} & - & \text{C} & - & \text{C} & - \text{H} \\  &   & &   & &   & \\  & \text{H} & & \text{H} & & \text{H} &   \end{array}  $ <p>propane</p>	5. $  \begin{array}{ccccccc}  & \text{H} & & \text{H} & & \text{H} & \\  &   & &   & &   & \\  \text{H} & - \text{C} & - & \text{C} & - & \text{C} & - \text{H} \\  &   & &   & &   & \\  & \text{H} & & \text{H} & & \text{H} & - \text{C} - \text{H} \\  & & & & & &   \\  & & & & & & \text{H}  \end{array}  $ <p>butane</p>
2. $  \begin{array}{ccccccc}  & \text{H} & & \text{H} & & \text{H} & & \text{H} \\  &   & &   & &   & &   \\  \text{H} & - \text{C} & = & \text{C} & - & \text{C} & - & \text{C} - \text{H} \\  & & & & &   & &   \\  & & & & & \text{H} & & \text{H}  \end{array}  $ <p>1-butene</p>	6. $  \begin{array}{ccccccc}  & \text{H} & & \text{CH}_3 & & \text{H} & \\  &   & &   & &   & \\  \text{H} & - \text{C} & - & \text{C} & - & \text{C} & - \text{H} \\  &   & &   & &   & \\  & \text{H} & & \text{H} & & \text{H} &   \end{array}  $ <p>2-methylpropane</p>
3. $  \begin{array}{ccccccc}  & & & & & \text{H} & \\  & & & & &   & \\  \text{H} & - \text{C} & \equiv & \text{C} & - & \text{C} & - \text{H} \\  & & & & &   & \\  & & & & & \text{H} &   \end{array}  $ <p>1-propyne</p>	7. $  \begin{array}{ccccccc}  & \text{H} & & \text{H} & & \text{H} & & \text{H} & & \text{H} \\  &   & &   & &   & &   & &   \\  \text{H} & - \text{C} & - & \text{C} & = & \text{C} & - & \text{C} & - & \text{C} - \text{H} \\  & & & & & & &   & &   \\  & & & & & & & \text{H} & & \text{H}  \end{array}  $ <p>2-pentene</p>
4. $  \begin{array}{ccccccc}  & \text{H} & & \text{H} & & \text{H} & & \text{CH}_3 & & \text{H} \\  &   & &   & &   & &   & &   \\  \text{H} & - \text{C} & - & \text{C} & - & \text{C} & - & \text{C} & - & \text{C} - \text{H} \\  &   & &   & &   & &   & &   \\  & \text{H} & & \text{H} & & \text{H} & & \text{H} & & \text{H}  \end{array}  $ <p>2-methylpentane</p>	8. $  \begin{array}{ccccccc}  & & & & & \text{H} & \\  & & & & &   & \\  & & & & & \text{H} - \text{C} - \text{H} & \\  & & & & &   & \\  & & & & & \text{H} - \text{C} - \text{H} & \\  & & & & &   & \\  & & & & & \text{C} & \\  & & & & &   & \\  & & & & & \text{H} - \text{C} - \text{H} & \\  & & & & &   & \\  & & & & & \text{H} & \\  & & & & &   & \\  & & & & & \text{H} - \text{C} - \text{H} & \\  & & & & &   & \\  & & & & & \text{H} & \\  & & & & &   & \\  & & & & & \text{H} - \text{C} - \text{H} & \\  & & & & &   & \\  & & & & & \text{H} &   \end{array}  $ <p>3,3-diethylpentane</p>

# Teaching Activity Naming And Creating Hydrocarbons Answer

**Oswaal Editorial Board**



## Teaching Activity Naming And Creating Hydrocarbons Answer:

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