



# Oxidation-Reduction Reactions

- Electrons can be neither created out of nothing nor destroyed
- In any redox reaction there is an element being reduced and an element being oxidized
- The total increase in the oxidation numbers must equal the total decrease in the oxidation numbers

If an element is reduced

- It gains electrons
- Oxidation number decreases
- The substance is classified as an oxidizing agent

If an element is oxidized

- It loses electrons
- Oxidation number increases
- The substance is classified as a reducing agent

# Teaching Transparency 56 Oxidation And Reduction Reactions

**Mike Jess**



## Teaching Transparency 56 Oxidation And Reduction Reactions:

Instructors Guide to Media and Print Resources McMurry, DonnaJean Fredeen, Robert Pribush, 1998-01-30

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**Boston Journal of Chemistry** ,1871 *Chemical Age* ,1928 **The Chemical Age** ,1928 *Index to 35mm Educational Filmstrips* National Information Center for Educational Media, 1975 *Boston Journal of Chemistry and Popular Science Review* ,1871 **Modern Inorganic Synthetic Chemistry** Ruren Xu, Yan Xu, 2017-02-11 Modern Inorganic Synthetic Chemistry Second Edition captures in five distinct sections the latest advancements in inorganic synthetic chemistry providing materials chemists chemical engineers and materials scientists with a valuable reference source to help them advance their research efforts and achieve breakthroughs Section one includes six chapters centering on synthetic chemistry under specific conditions such as high temperature low temperature and cryogenic hydrothermal and solvothermal high pressure photochemical and fusion conditions Section two focuses on the synthesis and related chemistry problems of highly distinct categories of inorganic compounds including superheavy elements coordination compounds and coordination polymers cluster compounds organometallic compounds inorganic polymers and nonstoichiometric compounds Section three elaborates on the synthetic chemistry of five important classes of inorganic functional materials namely ordered porous materials carbon materials advanced ceramic materials host guest materials and hierarchically structured materials Section four consists of four chapters where the synthesis of functional inorganic aggregates is discussed giving special attention to the growth of single crystals assembly of nanomaterials and preparation of amorphous materials and membranes The new edition s biggest highlight is Section five where the frontier in inorganic synthetic chemistry is reviewed by focusing on biomimetic synthesis and rationally designed synthesis Focuses on the chemistry of inorganic synthesis assembly and organization of wide ranging inorganic systems Covers all major methodologies of inorganic synthesis Provides state of the

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