

Section 8.1: Team Learning Worksheet

1. If a substance is insoluble, does this mean it is a non-electrolyte? If a substance is a non-electrolyte, does this mean it is insoluble? Explain.
2. Consider mixing aqueous solutions of lead(II) nitrate and sodium iodide.
 - a. Name the possible products and determine their formulas. Explain.
 - b. What is the product of this reaction? How do you know?
3. Why is the formation of a solid evidence of a chemical reaction? Use molecular-level sketches in your explanation.
4. If a soluble ionic compound is added to water, what does the resulting solution look like in water at a highly magnified level? Draw a comparable picture when an insoluble ionic compound is added to water.

Team Learning Worksheet 8 Scarsdale Answers

RJ Alexander



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