

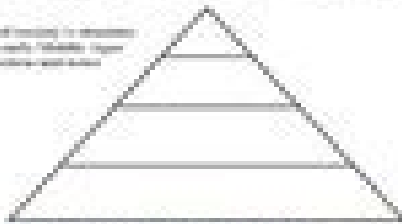
# 9 ENGAGING PAGES ON EVERY ASPECT OF THE UNIT!

## Key Concepts

Remember: The middle ages are often called the "middle ages" or "medieval ages" because they are in the middle of the history of the world.

### The Middle Ages

Complete the graphic representation of history in the middle ages. Write the names of the people who lived in the middle ages. Write the names of the people who lived in the middle ages. Write the names of the people who lived in the middle ages.



## The Crusades: Goal and Effect

The crusades were a series of religious wars.

GOAL: to reclaim	GOAL: to reclaim	GOAL: to reclaim	GOAL: to reclaim

What were the goals of the crusades? Write the names of the people who lived in the middle ages. Write the names of the people who lived in the middle ages. Write the names of the people who lived in the middle ages.

The crusades were a series of religious wars. Write the names of the people who lived in the middle ages. Write the names of the people who lived in the middle ages. Write the names of the people who lived in the middle ages.

What were the goals of the crusades?

What were the goals of the crusades? Write the names of the people who lived in the middle ages. Write the names of the people who lived in the middle ages. Write the names of the people who lived in the middle ages.

## Crusades in the Middle Ages

The crusades were a series of religious wars. Write the names of the people who lived in the middle ages. Write the names of the people who lived in the middle ages. Write the names of the people who lived in the middle ages.

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The crusades were a series of religious wars. Write the names of the people who lived in the middle ages. Write the names of the people who lived in the middle ages. Write the names of the people who lived in the middle ages.

## The Middle Ages in Europe Complete Unit Guide Packet

### Overview

The Middle Ages began around 500 AD and ended around 1500 AD. The Middle Ages were a time of great change and growth in Europe. The Middle Ages were a time of great change and growth in Europe.

The Middle Ages were a time of great change and growth in Europe. The Middle Ages were a time of great change and growth in Europe. The Middle Ages were a time of great change and growth in Europe.

The Middle Ages were a time of great change and growth in Europe. The Middle Ages were a time of great change and growth in Europe. The Middle Ages were a time of great change and growth in Europe.



## Geography of Europe in the Middle Ages

Remember: The Middle Ages were a time of great change and growth in Europe. The Middle Ages were a time of great change and growth in Europe. The Middle Ages were a time of great change and growth in Europe.



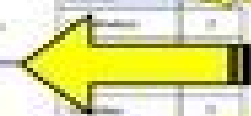
### European Land Connections

Spain	France	Germany
Italy	England	Sweden
Poland	Denmark	Netherlands
Belgium	Switzerland	Austria

### European Sea Connections

Spain	France
Italy	England
Poland	Denmark
Belgium	Switzerland

Intro & Essential Questions for the entire unit!



Question	Answer
What were the goals of the crusades?	
What were the goals of the crusades?	
What were the goals of the crusades?	
What were the goals of the crusades?	
What were the goals of the crusades?	

### Overview

The Middle Ages were a time of great change and growth in Europe. The Middle Ages were a time of great change and growth in Europe. The Middle Ages were a time of great change and growth in Europe.

# World History Middle Ages Study Guide Answers

**Thomas Griffiths**



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web with a strong acid and strong base titration strong acids and strong bases refer to species that completely dissociate to form ions in solution by contrast weak acids and bases ionize only partially and the ionization reaction is reversible

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web conclusion the purpose of the experiment was to identify the unknown solid acid 17 by calculating its molar mass this was done by titration of the acid with sodium hydroxide in order to find the volume of sodium hydroxide required to reach the end point and that value was used in unit analysis to find the molar mass and identify the unknown solid acid

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web chem 1210 spring 2019 experiment 10.11 part 1 acid base titration abstract the purpose of this experiment is to observe the titration of hydrochloric acid a strong acid with sodium hydroxide a strong base and acetic acid a

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web goals understand that parts of an acid base titration be able to determine the  $K_a$  or  $K_b$  from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate  $K_{a1}$  and  $K_{a2}$  for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1:1  $HCl(aq) + NaOH(aq) \rightarrow NaCl(aq) + H_2O(l)$  one mole of HCl would be fully neutralized by one mole of NaOH

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide NaOH and oxalic acid  $H_2C_2O_4 \cdot 2H_2O$  and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web acid base reactions can be used to determine concentrations of an acid or a base using stoichiometric calculations and titrimetric analysis the volume of a solution required to react

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web jul 19 2023 introduction titration is an analytical quantitative technique used to determine the concentration of a solute a ph titration is used to determine the concentration of an acid or a base titrations play an important role in determining amount and purity in many manufacturing processes

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pka of an unknown acid or pkb of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance  $\text{H}_2\text{C}_2\text{O}_4$  aq  $2\text{NaOH}$  aq  $\text{Na}_2\text{C}_2\text{O}_4$   $4\text{H}_2\text{O}$  to differentiate an acid from a base in a solution a ph scale is used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction  $\text{NaOH}$  aq  $\text{C}_6\text{H}_4\text{COOH}$  aq  $\text{C}_6\text{H}_4\text{COONa}$  aq  $\text{H}_2\text{O}$  l the point of equivalence is the point of titration when the number of moles of naoh na added is equal to the number of moles of khp nb in the

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