

List of Common Polyatomic Ions

Cations:

<u>Symbol</u>	<u>Name</u>	<u>Charge</u>	<u>Symbol</u>	<u>Name</u>	<u>Charge</u>
NH_4^+	Ammonium	+1	Hg_2^{+2}	Dimercury (I)	+2

Anions:

<u>Symbol</u>	<u>Name</u>	<u>Charge</u>	<u>Symbol</u>	<u>Name</u>	<u>Charge</u>
$\text{C}_2\text{H}_3\text{O}_2^-$	Acetate	-1	CO_3^{-2}	Carbonate	-2
HCO_3^-	Bicarbonate	-1	CrO_4^{-2}	Chromate	-2
BrO_3^-	Bromate	-1	$\text{Cr}_2\text{O}_7^{-2}$	Dichromate	-2
BrO_2^-	Bromite	-1	$\text{C}_2\text{O}_4^{-2}$	Oxalate	-2
ClO_4^-	Perchlorate	-1	O_2^{-2}	Peroxide	-2
ClO_3^-	Chlorate	-1	SO_4^{-2}	Sulfate	-2
ClO_2^-	Chlorite	-1	SO_3^{-2}	Sulfite	-2
ClO^-	Hypochlorite	-1	$\text{S}_2\text{O}_3^{-2}$	Thiosulfate	-2
CN^-	Cyanide	-1	SiO_3^{-2}	Silicate	-2
SCN^-	Thiocyanate	-1	MoO_4^{-2}	Molybdate	-2
OH^-	Hydroxide	-1	WO_4^{-2}	Tungstate	-2
NO_3^-	Nitrate	-1	BO_3^{-3}	Borate	-3
NO_2^-	Nitrite	-1	PO_4^{-3}	Phosphate	-3
MnO_4^-	Permanganate	-1	PO_3^{-3}	Phosphite	-3
HSO_4^-	Bisulfate	-1	AsO_4^{-3}	Arsenate	-3
HSO_3^-	Bisulfite	-1	AsO_3^{-3}	Arsenite	-3
IO_3^-	Iodate	-1			
IO_4^-	Periodate	-1			
$\text{C}_6\text{H}_5\text{COO}^-$	Benzoate	-1			

<u>Symbol</u>	<u>Name</u>	<u>Charge</u>
$\text{H}_2\text{PO}_4^{-1}$	Dihydrogen phosphate	-1
HPO_4^{-2}	Hydrogen phosphate	-2
$\text{Fe}(\text{CN})_6^{-3}$	Hexacyanoferrate (III)	-3
$\text{Fe}(\text{CN})_6^{-4}$	Hexacyanoferrate (II)	-4

Polyatomic Ions Extension Questions Answers

United States. Office of Saline Water

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the text 'United States. Office of Saline Water'.

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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 *Chemistry The Molecular Nature of Matter* 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

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the whole period and some evening sessions where the participants were offered the opportunity to present their own work and discuss their ideas with senior scientists. Since the Institute took place a few weeks after Prof Dr John C Slater was suddenly taken from our scientific community it was a great honor for us to dedicate this course on behalf of the organizing committee to the late John C Slater.

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