

Ch 13 Study Guide (Chemical Bonds)

1. Fill in the table below with the correct values:

Element Name	Ion Symbol	# Protons	# Electrons	Charge	Ion Type
Fluorine	F^-	9	10	-1	Anion
Iodine	I^-	53	54	-1	Anion
Sulfur	S^{2-}	16	18	-2	Anion
Potassium	K^+	19	18	+1	Cation
Calcium	Ca^{2+}	20	18	+2	Cation
Bromine	Br^-	35	36	-1	Anion
Strontium	Sr^{2+}	38	36	+2	Cation
Oxygen	O^{2-}	8	10	-2	Anion
Magnesium	Mg^{2+}	12	10	+2	Cation
Aluminum	Al^{3+}	13	10	+3	Cation
Selenium	Se^{2-}	34	36	-2	Anion
Lithium	Li^+	3	2	+1	Cation
Rubidium	Rb^+	37	36	+1	Cation
Chlorine	Cl^-	17	18	-1	Anion

2. An ionic bond is formed between what two types of atoms?

Metal & nonmetal

3. A covalent bond is formed between what two types of atoms?

Nonmetal & Nonmetal

4. Draw the bond(s) that will form between Mg and O.



5. What is the formula for the compound formed in question #4?



Valence Electrons Chemistry If8766 Answers

Ian Pickup



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Valence and the Structure of Atoms and Molecules Gilbert Newton Lewis (Chemiker, USA), 1923 *A Study of Valence with Reference to Secondary School Chemistry* Mary Alma Rose Gaffney, 1958 The purpose of this study is to investigate the place of valence as a topic in high school chemistry with a view to finding possible means of closing the gap between what is actually achieved at the secondary level and what seems to be a desirable point as which to initiate college study Some of the factors responsible for this might well be the confusion in terminology the lag between the enunciation of theories and their appearance in text books the difficulty of symbolic representation of atomic and molecular structure and the closed mind many students develop toward change The extent to which these difficulties can be resolved in the light of research studies historical development of the theory future work to be encountered and the level of maturity of the high school pupil will be

considered Historical works surveys of educational research scientific and educational journals text books courses of studies and works on methodology have been consulted The Introduction and Statement of the Problem 1 2 3 *Roadmap to the Virginia SOL* Princeton Review,2005 Roadmap to the Virginia SOL EOC Chemistry includes strategies that are proven to enhance student performance The experts at The Princeton Review provide content review of the crucial material most likely to appear on the test detailed lessons complete with test taking techniques for improving test scores 2 complete practice Virginia SOL EOC Chemistry tests **Valence and the structure of atoms and molecules** Gilbert N. Lewis,1966 All
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