

Valence Electron Practice

Name: _____ Period: _____

1. What is a valence electron and why are they important to a chemist?

Valence electrons are used in bonding and determine properties/characteristics of elements

2. How can you find the number of valence electrons an element has?

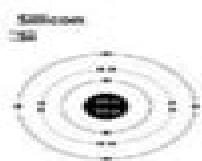
The group number on the periodic table tells you the number of valence electrons an element has.

3. How many protons are in the following elements?

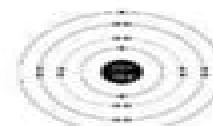
A. Carbon B. Hydrogen C. Oxygen D. Phosphorus E. Calcium
6 1 8 15 20

4. Draw the atomic structures for the following. Include the number of protons, neutrons and draw in the valence electrons.

A. Silicon



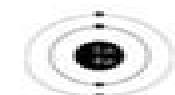
B. Potassium



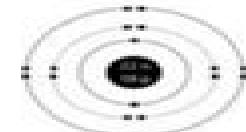
C. Sulfur



D. Beryllium



E. Argon



F. Nitrogen



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