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Chemistry Paper 2 (Practical), WASSCE (SC), 2022

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Question 1

- (a) (i) Define an acid according to the Lewis concept.
 (ii) Give one example of a Lewis acid. [3 marks]
- (b) Explain salting out in soap preparation. [2 marks]
- (c) State the reagent and condition necessary for the following conversion:

$$\text{H-C}\equiv\text{C-H} \rightarrow \text{Ag} - \text{C}\equiv\text{C} - \text{Ag}$$
 [2 marks]
- (d) What is the percentage abundance of an isotope? [2 marks]
- (e) (i) Why does the element with atomic number 18 not have an oxide?
 (ii) Explain why chlorine(I) oxide has a low melting point. [3 marks]
- (f) Describe a test to distinguish between concentrated HNO_3 and concentrated H_2SO_4 . [3 marks]
- (g) State two differences between an *electrochemical cell* and an *electrolytic cell*. [2 marks]
- (h) How does the trend in ionization energy affect the reactivity of group I element? [3 marks]
 Define the term *molecular formula*

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