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Date: December 9, 2020

Student Exploration: Feel the Heat

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: calorimeter, conductor, controlled experiment, dissolve, endothermic, exothermic, insulator, solute, solution, solvent, surroundings, system

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Eduardo hurts his knee during a basketball game. The trainer applies a cold pack, which gets cold after being squeezed. The next day, Eduardo's friend Beth is going sledding. Beth buys some hand warmers and puts them into her gloves. All morning, her hands stay toasty warm.

1. How do you think these devices work?

I think you can control it by temperature .

2. Where do you think the "cold" and the heat comes from?

I think it comes from the friction you make from both of your hands warmer and cold packs .

Gizmo Warm-up

As you saw with the instant cold pack and the hand warmers, some reactions absorb energy while others release it. In the *Feel the Heat* Gizmo, you will explore these energy changes while making your own hot and cold packs.

To begin, select the TEST POWDERS tab. Drag a bottle of $\text{NaC}_2\text{H}_3\text{O}_2$ (sodium acetate) from the shelf. Using the sliders, decide how much water and powder to add. Note the beginning temperature, and then press **Play** (▶) to see what happens.



1. Did the temperature go up or down?

The temperature went up .

2. Experiment with different powders until you find one that produces the opposite effect.

What powder did you use?

I used NH_4NO_3 ammonium nitrate .

When a powder dissolves in water a **solution** is formed. The powder is the **solute** and the water is the **solvent**. Oftentimes energy changes accompany the formation of a solution. When added to water, some powders cause the resulting solution to get hot, while others make it cold.

Options For Youth Answer Key

Michael Seilmaier

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