

Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question. using the letters in each "box" to identify the required element(s).

A																				B
C																				
	F						G								H					K
	L						M										I			
J		Z					M					N					O	P		
Q		R		S	T	U									V					
			V									W								
		X										Y								

List elements in alphabetical order when more than 1 element answers a question.

- 1) Element A is in a box all by itself because: A) it is really a Noble Gas B) it is a gas C) it is NOT a gas D) it does NOT share the properties of the other elements in its group E) it is a liquid
- 2) Which of these elements is/are highly reactive? A) "C" B) "S" C) "X" D) A, B, and C E) A and B only
- 3) What is the letter of the element whose atomic number is closest to the element labeled "Z"?
- 4) Which letter labels the halogen element with the lowest atomic number?
- 5) Which of these elements is the least reactive? A) "Q" B) "Y" C) "A" D) "O" E) "R"
- 6) Which of these elements is/are NOT transition elements/metals (list ALL that are not transition elements)? A) "G" B) "N" C) "M" D) "H" E) "V"

continued on other side

Section 6 1 Development Modern Periodic Table

Worksheet Answers

David Baud

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