



Reinforcement

Chemical Changes

Read the passage and answer the following questions:



1. What are the products of this chemical reaction?

As mentioned in the introduction, the burning of methane in this chemical reaction, the products are water and carbon dioxide. The reaction is exothermic, meaning it releases heat. The reaction is also a combustion reaction, meaning it involves the burning of a substance.

2. What is the reactant in this chemical reaction? Is it a solid, liquid, gas, or plasma? Is it a pure substance or a mixture?

3. What is the physical state of the reactant and the products?

The reactant, hydrogen peroxide, is a colorless liquid. The products, water and oxygen, are also colorless. Water is a liquid at room temperature, while oxygen is a gas. The reaction is exothermic, meaning it releases heat. The reaction is also a combustion reaction, meaning it involves the burning of a substance.

4. According to the law of conservation of mass, if 100g of hydrogen peroxide decomposes, how much water and oxygen will be produced?

100 grams



5. What is the balanced chemical equation for this reaction? Is it?

6. What is the balanced chemical equation for this reaction? Is it?

7. What are the physical states of the reactant and the products?

Hydrogen peroxide is a liquid.
Water is a liquid.
Oxygen is a gas.

8. What is the physical state of the reactant and the products?

9. What is the balanced chemical equation for this reaction? Is it?

The reaction is exothermic, meaning it releases heat. The reaction is also a combustion reaction, meaning it involves the burning of a substance.

10. According to the law of conservation of mass, if 100g of hydrogen peroxide decomposes, how much water and oxygen will be produced?

Section 3 Reinforcement Physical And Chemical Changes Answers

**George Frederick Shrady, Thomas
Lathrop Stedman**



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Holt Science and Technology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001-07 *Addison-Wesley Introduction to Physical Science* Michael B. Leyden, 1988 *Matter, Building Block of the Universe*, 1993 *Chemistry* George Tyler Miller, 1987 **Matter** Prentice-Hall Staff, 1994 *Physical Science Anted, Grades 6-9* David W. Appenbrink, 1986 **Foundations of Physical Science** Tom Hsu PH.D., 2002 **Resources for Teaching Middle School Science** Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents **Nature Science** Prentice-Hall Staff, 1994 *Chemical Matter* Prentice-Hall Staff, 1994 Authorized teaching resource in Alberta for senior

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