Quiz & Worksheet - Radioactive Nuclei & Decay

Which type of decay will a sample of Radium-223 most likely undergo?

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				800	=	=	18.0	40	20	Acti	=	807	77	=	100	100	Section 1	E

- Alpha
- Beca
- Gamma
- It is a sable isotope and will not undergo nuclear decay
- 2. Which type of radioactive decay will a sample of Carbon-14 undergo?

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20	34	200		-53	42	40	-	-		407 (Fig.	-00	=		200	2	50.0	1
509 Ca	104			70	2	2	79.	200	2	70	-	**	-	800	=	-	=
-	20			Dame.	700	200	200	-	Day.	No.	STEEP COm-	SAM.	114	112	200	STORE	2.1
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			=		-	70.0	20	20	Marin I	=	-	=	70	100	200	100	18

- Alpha
- Besa
- Gamma
- It is a sable isotope and will not undergo nuclear decay
- 3. Please utilize the provided equations to answer the following radioactive decay question.

A radioactive sample has a half-life of 5 seconds. The sample confians 5.0×10^{27} radioactive aforms. How many radioactive atoms will be left after 30 seconds? The necessary equations are listed below.

$$N = N_0 e^{-3\epsilon}$$

$$\lambda = \frac{\ln (2)}{\epsilon_b}$$

- 5.0 x 10³⁰ atoms
- 2.8 x 1032 atoms
- 2.6 x 10³⁴ atoms
- 6.0 x 10³² atoms

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