Name	Class	Date
Chapter C. Chapter of Balada		

Section 6.2 Covalent Bonding September 1465-14601

This section discusses the formation of construt bonds and the factors that determine whether a molecule is polar or mangolar. It also discusses attenutions Persons makes les-

Reading Strategy (2000 195)

Relating Test and Visuals. As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of resource beautifurerate.

Materials Works		
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Direction det	2000000	
Obuctural formula		
Space-filling		
Dection claud		

Covalent Bonds were 195-1971

- Describe a covalent bond.
- 2. Circle the letters of prolocular models that show orbitals of atoms overlapping when a covalent bond forms.
 - a, electron det

b. structural foresula-

c. space-filling

d. electron cloud

- Describe a molecule.
- 4. Is the following sentence tree or false? In a covalent bond, the atomsare held together by the attractions between the shared electrons and the protons in each murleus.
- 5. Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
 - And interested

the comment

C. Davido

ell. Histories

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Yijin Wang

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