

Name Solutions

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### Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



| Atoms | Valence electrons | Electron transfer from/to each atom | Ions formed in the product |
|-------|-------------------|-------------------------------------|----------------------------|
| Li    | 1 ve              | transfers 1ve <sup>-</sup> to F     | Li <sup>+</sup>            |
| F     | 7 ve              | accepts 1ve <sup>-</sup> from Li    | F <sup>-</sup>             |

| Reactions                                              | Atoms | Valence electrons | Electron transfer from/to each atom | Ions formed in the product |
|--------------------------------------------------------|-------|-------------------|-------------------------------------|----------------------------|
| 1) $\text{Li} + \text{Cl} \Rightarrow \text{LiCl}$<br> | Li    | 1 ve              | donate 1e <sup>-</sup>              | Li <sup>+</sup>            |
|                                                        | Cl    | 1 ve              | accepts 1e <sup>-</sup>             | Cl <sup>-</sup>            |
| 2) $\text{Ca} + \text{O} \Rightarrow \text{CaO}$<br>   | Ca    | 2 ve              | donates 2e <sup>-</sup>             | Ca <sup>+2</sup>           |
|                                                        | O     | 6 ve              | accepts 2e <sup>-</sup>             | O <sup>-2</sup>            |
| 3) $\text{Be} + \text{F} \Rightarrow \text{BeF}_2$<br> | Be    | 2 ve              | donates 2e <sup>-</sup> to each F   | Be <sup>+2</sup>           |
|                                                        | F     | 7 ve              | each F accepts 1ve each             | F <sup>-</sup>             |
| 4) $\text{Mg} + \text{S} \Rightarrow \text{MgS}$<br>   | Mg    | 2 ve              | donates 2e <sup>-</sup> to S        | Mg <sup>+2</sup>           |
|                                                        | S     | 6 ve              | accepts 2e <sup>-</sup>             | S <sup>-2</sup>            |
| 5) $\text{K} + \text{F} \Rightarrow \text{KF}$<br>     | K     | 1 ve              | donates 1ve <sup>-</sup>            | K <sup>+1</sup>            |
|                                                        | F     | 7 ve              | accepts 1ve <sup>-</sup>            | F <sup>-1</sup>            |

# Section 6 2 Ionic Bonding Worksheet Answer Key

**IM Harris**



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