



COVALENT BONDING

Practice Problems

Do your best to solve each of the following problems.

SECTION 8.1 MOLECULAR COMPOUNDS

1. Classify each of the following compounds as ionic or molecular.

- a. Na_2SO_4 b. CH_3NH_2 c. HCl d. H_2O e. H_2O_2 f. H_2O_2 g. H_2O_2 h. H_2O_2 i. H_2O_2 j. H_2O_2 k. H_2O_2 l. H_2O_2 m. H_2O_2 n. H_2O_2 o. H_2O_2 p. H_2O_2 q. H_2O_2 r. H_2O_2 s. H_2O_2 t. H_2O_2 u. H_2O_2 v. H_2O_2 w. H_2O_2 x. H_2O_2 y. H_2O_2 z. H_2O_2

2. Write the formula for the following molecular compounds.

- a. H_2O b. H_2O_2 c. H_2O_2 d. H_2O_2 e. H_2O_2 f. H_2O_2 g. H_2O_2 h. H_2O_2 i. H_2O_2 j. H_2O_2 k. H_2O_2 l. H_2O_2 m. H_2O_2 n. H_2O_2 o. H_2O_2 p. H_2O_2 q. H_2O_2 r. H_2O_2 s. H_2O_2 t. H_2O_2 u. H_2O_2 v. H_2O_2 w. H_2O_2 x. H_2O_2 y. H_2O_2 z. H_2O_2

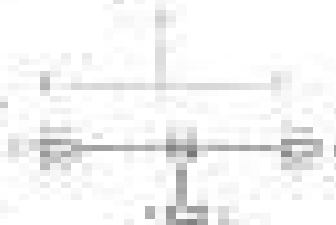
3. Write the formula for the following molecular compounds. (Note: The following compounds are not molecular compounds.)
4. Write the formula for the following molecular compounds. (Note: The following compounds are not molecular compounds.)
5. Write the formula for the following molecular compounds. (Note: The following compounds are not molecular compounds.)

SECTION 8.2 THE NATURE OF COVALENT BONDING

1. Draw the Lewis structure for the following molecule.



2. Draw the Lewis structure for the following molecule.



3. Draw the Lewis structure for the following molecule.



4. Draw the Lewis structure for the following molecule.

5. Draw the Lewis structure for the following molecule. (Note: The following compound is not a molecular compound.)



6. Draw the Lewis structure for the following molecule.

SECTION 8.3 BONDING: THE ORBITAL

1. Write the formula for the following molecular compounds. (Note: The following compounds are not molecular compounds.)
2. Write the formula for the following molecular compounds. (Note: The following compounds are not molecular compounds.)

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JL Elias



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