



*Figure 2 - Controls on the top of the D5100*

# Nikon D5100 Manual Focus Lenses

**Tobias Bleicker**



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web the relative amounts of products and reactants at equilibrium is characterized by the equilibrium constant for this reaction  $K_{eq}$  is defined as  $\frac{[enol]_{eq}}{[keto]_{eq}}$  where the subscript eq indicates that all concentrations are measured at equilibrium

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web using NMR spectroscopy the equilibrium values for different ketones which undergo tautomerization were found by determining relative peak values for each species the value for acetylacetone was found to be  $k_{\text{c}}/k_{\text{k}} = 6.622$  and for ethyl acetoacetate was  $k_{\text{c}}/k_{\text{k}} = 7.510$

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