natural transmutation - Uranium spenteneously decays.

$$^{238}_{92}U \rightarrow ^{234}_{90}Th + ^{4}_{2}He$$

artificial transmutation - bombardment of a stable isotope to force it to decay.

$${}_{7}^{14}N + {}_{2}^{4}Hg \rightarrow {}_{1}^{1}p +$$

When the bullets are \_\_\_\_ charged, they are \_\_\_\_\_ by the nucleus they are benbarding. To overcome the repulsions, they must be \_\_\_\_\_ to very high speeds by \_\_\_\_\_ possionstors.

nuclear fission - Heavy nuclei are bembarded with neutrons and split.

$$_{0}^{1}n + {}_{92}^{235}U \rightarrow {}_{56}^{142}Ba + {}_{36}^{91}Kr + 3({}_{0}^{1}n) + E$$

Mass of particles produced is slightly \_\_\_\_\_ than the mass of the reactants.

This mass is converted into \_\_\_\_ (E = \_\_\_)

critical mass: \_\_\_\_\_\_ mass of \_\_\_\_\_ material required

for a

nuclear reactors control fission \_\_\_\_\_\_ reactions to produce energy dangers

nuclear fusion - combination of \_\_\_\_\_ nuclei into \_\_\_\_ with release of

$${}_{1}^{2}H + {}_{1}^{3}H \rightarrow {}_{2}^{4}He + {}_{0}^{1}n + E$$

Mass of particles produced is much \_\_\_\_\_ than the mass of the \_\_\_\_\_ This \_\_\_\_\_ is converted into energy. (E + \_\_\_\_\_)

On back, fot advantages of and problems with using fusion as an energy source.

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Michael Rajnik

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