

Haloalkanes Hydrogen atom in aliphatic hydrocarbons replaced by halogen. $R-X$ ($X = F, Cl, Br, I$)

Classification:

- Haloalkanes
 - 1° $R-CH_2-X$
 - 2° R_2-CH-X
 - 3° R_3-C-X

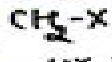
Allylic Halide



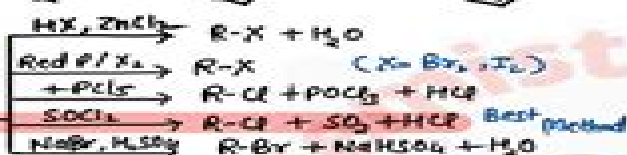
Benzyl Halide

Vinyl Halide

Aryl Halide



Preparation



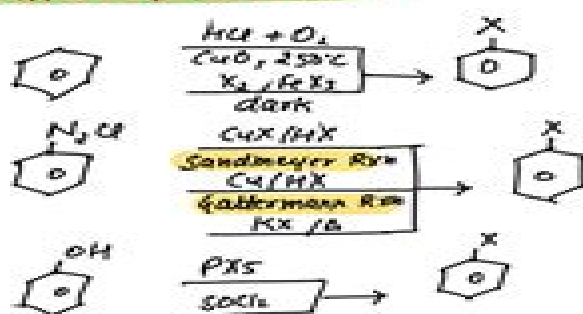
From Alkene



Halogenation:



Preparation of Haloarenes



Physical Properties

- B.P \propto Mol. mass
- B.P \propto Surface Area
- B.P \propto 1/ Branching

- B.Pt Order $\rightarrow R-I > R-Br > R-Cl > R-F$
- M.Pt of p-isomer is more than that of ortho and meta due to symmetry.
- The density increases with increase in no. of carbon atoms, halogen atoms and atomic mass of the halogen atom.
- Solubility** Although Haloalkanes are polar but they can not form Hydrogen bond with water molecule hence they are insoluble in water and soluble in organic solvents.

Finkelstein Reaction



Swarts Reduction



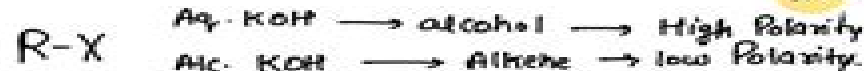
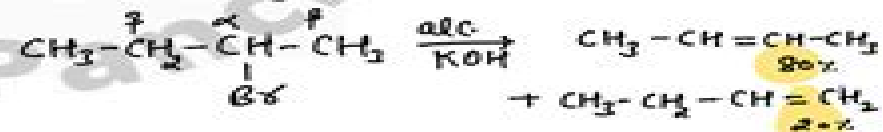
Chemical Properties

Elimination Reaction

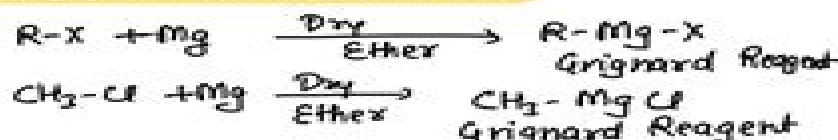


Acc. to Saytzeff Rule

In elimination reaction, preferred alkene is one in which double bonded C-atom are more alkylated



Reaction with Metal



Grignard reagent are highly reactive and react with any source of proton to give hydrocarbons

Notes On Organic Chemistry 1 Chemistry 3 Klb

Xiang Xie



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Table of Contents Notes On Organic Chemistry 1 Chemistry 3 Klb

1. Understanding the eBook Notes On Organic Chemistry 1 Chemistry 3 Klb
 - The Rise of Digital Reading Notes On Organic Chemistry 1 Chemistry 3 Klb
 - Advantages of eBooks Over Traditional Books
2. Identifying Notes On Organic Chemistry 1 Chemistry 3 Klb
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Notes On Organic Chemistry 1 Chemistry 3 Klb
 - User-Friendly Interface
4. Exploring eBook Recommendations from Notes On Organic Chemistry 1 Chemistry 3 Klb
 - Personalized Recommendations
 - Notes On Organic Chemistry 1 Chemistry 3 Klb User Reviews and Ratings
 - Notes On Organic Chemistry 1 Chemistry 3 Klb and Bestseller Lists
5. Accessing Notes On Organic Chemistry 1 Chemistry 3 Klb Free and Paid eBooks
 - Notes On Organic Chemistry 1 Chemistry 3 Klb Public Domain eBooks
 - Notes On Organic Chemistry 1 Chemistry 3 Klb eBook Subscription Services
 - Notes On Organic Chemistry 1 Chemistry 3 Klb Budget-Friendly Options
6. Navigating Notes On Organic Chemistry 1 Chemistry 3 Klb eBook Formats

- ePub, PDF, MOBI, and More
 - Notes On Organic Chemistry 1 Chemistry 3 Klb Compatibility with Devices
 - Notes On Organic Chemistry 1 Chemistry 3 Klb Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Notes On Organic Chemistry 1 Chemistry 3 Klb
 - Highlighting and Note-Taking Notes On Organic Chemistry 1 Chemistry 3 Klb
 - Interactive Elements Notes On Organic Chemistry 1 Chemistry 3 Klb
 8. Staying Engaged with Notes On Organic Chemistry 1 Chemistry 3 Klb
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Notes On Organic Chemistry 1 Chemistry 3 Klb
 9. Balancing eBooks and Physical Books Notes On Organic Chemistry 1 Chemistry 3 Klb
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Notes On Organic Chemistry 1 Chemistry 3 Klb
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Notes On Organic Chemistry 1 Chemistry 3 Klb
 - Setting Reading Goals Notes On Organic Chemistry 1 Chemistry 3 Klb
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Notes On Organic Chemistry 1 Chemistry 3 Klb
 - Fact-Checking eBook Content of Notes On Organic Chemistry 1 Chemistry 3 Klb
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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