

$$\Delta G^\circ = -2.86 \text{ kJ}$$

ह) मान निकाल E° की मान

$$\Delta G^\circ = -2.86 \text{ kJ}$$

$$\log K_c = \frac{-\Delta G^\circ}{2.303RT}$$

$$= \frac{(-2.86)}{2.303 \times 8.314 \times 298} = -0.5014$$

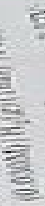
अतः

$$K_c = 10^{-0.5014}$$

$$K_c = 0.32$$

5. निम्नलिखित सेले को प्ले X पर सेल स्क्रिप्ट एवं and

निर्धार



दिए गए हैं

$$E^\circ_{\text{Mg}^{2+}/\text{Mg}} = -2.37 \text{ V}$$

$$E^\circ_{\text{Co}^{2+}/\text{Co}} = -0.28 \text{ V}$$

$$E^\circ_{\text{Fe}^{2+}/\text{Fe}} = -0.44 \text{ V}$$

$$E^\circ_{\text{Zn}^{2+}/\text{Zn}} = -0.76 \text{ V}$$

$$E^\circ_{\text{Pb}^{2+}/\text{Pb}} = -0.13 \text{ V}$$

$$E^\circ_{\text{H}^+/\text{H}_2} = 0 \text{ V}$$

$$E^\circ_{\text{Pt}} = 0 \text{ V}$$

$$E^\circ_{\text{H}_2/\text{Pt}} = 0 \text{ V}$$

$$E^\circ_{\text{H}^+/\text{H}_2} = 0 \text{ V}$$

$$E^\circ_{\text{H}_2/\text{Pt}} = 0 \text{ V}$$

$$E^\circ_{\text{H}^+/\text{H}_2} = 0 \text{ V}$$

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$$E^\circ_{\text{H}_2/\text{Pt}} = 0 \text{ V}$$

$$E^\circ_{\text{H}^+/\text{H}_2} = 0 \text{ V}$$

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1. Understanding the eBook Notes Chemistry For Up Board
 - The Rise of Digital Reading Notes Chemistry For Up Board
 - Advantages of eBooks Over Traditional Books
2. Identifying Notes Chemistry For Up Board
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Notes Chemistry For Up Board
 - User-Friendly Interface
4. Exploring eBook Recommendations from Notes Chemistry For Up Board
 - Personalized Recommendations
 - Notes Chemistry For Up Board User Reviews and Ratings
 - Notes Chemistry For Up Board and Bestseller Lists
5. Accessing Notes Chemistry For Up Board Free and Paid eBooks
 - Notes Chemistry For Up Board Public Domain eBooks
 - Notes Chemistry For Up Board eBook Subscription Services
 - Notes Chemistry For Up Board Budget-Friendly Options

6. Navigating Notes Chemistry For Up Board eBook Formats
 - ePub, PDF, MOBI, and More
 - Notes Chemistry For Up Board Compatibility with Devices
 - Notes Chemistry For Up Board Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Notes Chemistry For Up Board
 - Highlighting and Note-Taking Notes Chemistry For Up Board
 - Interactive Elements Notes Chemistry For Up Board
8. Staying Engaged with Notes Chemistry For Up Board
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Notes Chemistry For Up Board
9. Balancing eBooks and Physical Books Notes Chemistry For Up Board
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Notes Chemistry For Up Board
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Notes Chemistry For Up Board
 - Setting Reading Goals Notes Chemistry For Up Board
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Notes Chemistry For Up Board
 - Fact-Checking eBook Content of Notes Chemistry For Up Board
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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