

Physics

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(Chapter 1) (Units and Measurement) (Exercises)

(Class 11)

Question 1.14:

The unit of length convenient on the atomic scale is known as an angstrom and is denoted by Å: $1 \text{ Å} = 10^{-10} \text{ m}$. The size of a hydrogen atom is about 0.5 Å . What is the total atomic volume in m^3 of a mole of hydrogen atoms?

Answer 1.14:

Radius of hydrogen atom, $r = 0.5 \text{ Å} = 0.5 \times 10^{-10} \text{ m}$

Volume of hydrogen atom $= \frac{4}{3} \pi r^3$

$$= \frac{4}{3} \times \frac{22}{7} \times (0.5 \times 10^{-10})^3$$

$$= 0.524 \times 10^{-30} \text{ m}^3$$

1 mole of hydrogen contains 6.023×10^{23} hydrogen atoms.

Volume of 1 mole of hydrogen atoms

$$= 6.023 \times 10^{23} \times 0.524 \times 10^{-30}$$

$$= 3.16 \times 10^{-7} \text{ m}^3$$

Question 1.15:

One mole of an ideal gas at standard temperature and pressure occupies 22.4 L (molar volume). What is the ratio of molar volume to the atomic volume of a mole of hydrogen? (Take the size of hydrogen molecule to be about 1 Å). Why is this ratio so large?

Answer 1.15:

Radius of hydrogen atom, $r = 0.5 \text{ Å} = 0.5 \times 10^{-10} \text{ m}$

$$\text{Volume of hydrogen atom} = \frac{4}{3} \pi r^3 = \frac{4}{3} \times \frac{22}{7} \times (0.5 \times 10^{-10})^3 = 0.524 \times 10^{-30} \text{ m}^3$$

Now, 1 mole of hydrogen contains 6.023×10^{23} hydrogen atoms.

$$\therefore \text{Volume of 1 mole of hydrogen atoms, } V_a = 6.023 \times 10^{23} \times 0.524 \times 10^{-30}$$

$$= 3.16 \times 10^{-7} \text{ m}^3$$

Molar volume of 1 mole of hydrogen atoms at STP, $V_m = 22.4 \text{ L} = 22.4 \times 10^{-3} \text{ m}^3$

$$\therefore \frac{V_m}{V_a} = \frac{22.4 \times 10^{-3}}{3.16 \times 10^{-7}} = 7.08 \times 10^4$$

Hence, the molar volume is 7.08×10^4 times higher than the atomic volume. For this reason, the inter-atomic separation in hydrogen gas is much larger than the size of a hydrogen atom.

Question 1.16:

Explain this common observation clearly: If you look out of the window of a fast moving train, the nearby trees, houses etc. seem to move rapidly in a direction opposite to the train's motion, but the distant objects (hill tops, the Moon, the stars etc.) seem to be stationary. (In fact, since you are aware that you are moving, these distant objects seem to move with you).

Answer 1.16:

Line of sight is defined as an imaginary line joining an object and an observer's eye. When we observe nearby stationary objects such as trees, houses, etc. while sitting in a moving train, they appear to move rapidly in the opposite direction because the line of sight changes very rapidly.

On the other hand, distant objects such as trees, stars, etc. appear stationary because of the large distance. As a result, the line of sight does not change its direction rapidly.

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pedagogy and curriculum analyzes his conception of morals and religion to the extent that these ideas are specifically related to education examines his notions of educational psychology determines Lichtenberg's views on British education in the eighteenth century compares some of Lichtenberg's educational ideas in the works of contemporary thinkers and educators notably Schopenhauer James and Dewey A concomitant aspect of this book is a portrayal of Lichtenberg as found in his works viz as a student professor philosopher educator moralist psychologist comparative educationist as a searcher for absolute educational truth attainable only in a world to come SVEIN ØKSENHOLT PH D **A History of the Modern Fact** Mary Poovey, 2009-11-30 How did the fact become modernity's most favored unit of knowledge How did description come to seem separable from theory in the precursors of economics and the social sciences Mary Poovey explores these questions in A History of the Modern Fact ranging across an astonishing array of texts and ideas from the publication of the first British manual on double entry bookkeeping in 1588 to the institutionalization of statistics in the 1830s She shows how the production of systematic knowledge from descriptions of observed particulars influenced government how numerical representation became the privileged vehicle for generating useful facts and how belief whether figured as credit credibility or credulity remained essential to the production of knowledge Illuminating the epistemological conditions that have made modern social and economic knowledge possible A History of the Modern Fact provides important contributions to the history of political thought economics science and philosophy as well as to literary and cultural criticism Faith, Physics, and Psychology John Fitzgerald Medina, 2006 In Faith Physics and Psychology John Fitzgerald Medina offers a new understanding of the important role of religion and spirituality in the building of a global society **Concord Harvest**, 1970

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