ALKENES:

General formula = Cn.Hzn

functional group: c=C (savurated)

REACTIONS:

Hydration (420) => +-c-c-4

- accidit carryst
- nigh pressurer remp
- Forms account

Halogenarion(802) => 4 - 2-2-4

Hydrogenation(42) => # #

-nriker calonyst

composition =7 complete=(02th20 incomplete=10tH20

WITCOHOIT:

functional group on

General formua = Cattenn Ott

REACTIONS:

Burning => Product: CO2 +420

Add water (420) => (no H+ or OH-)

Add oxygen (02) => forms (almoxytic address

formentation => Ethanoll +1 Time . - filterto remove
yeast -disni remove water-

CABOXYLIC ACIDS:

functional group = COOH

General formula: (mHznerCOOH.)

REACTIONS:

Acid+Water => CH3(00H= CH3(00"+H" weak acid-partially lonised "

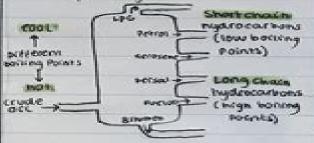
Acid+ (albonate => (H3(00H+ Nat(03 -> (0)+ H20+ CH3(00NA (ag) due to CO2

ACCO + ALCOHOT -> CH3COOH + CH3CHEOH -> CH3C COCH3CH2 + HILO ESTER.

eg emanoic acid+ erhanol > erhyl erhanoate.

FRACTIONAL DISTULATION:

vised to apperous musicocar bon fractions.



POLYMERISATION:

Addition Polymerisation - No side products

Condensation paymerisation . water side module

HO-E-O-E-OH + HO--OH =7 HO-E-O-D-O

20 modes: 40- 2-0-2-04 + H-N-1-1-4-4 (N-H-1-4H)

- [2-0-8-N-0-N-] +HEO

CRACKING

The breaking up the hydrocarbon chouns which are too long from fractional distribution into shorter More vietus hydrocarbons.

TWO Types:

O Steam cracking: Vaporised hydrocarbon + catalyst.

Example:

(8H18 =7 (1H6 46H14

one alkene

C+ HI4+CH4

one alkane.

(6 H12+ (2H6

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Nyenty Arrey

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Indian Journal of Chemistry ,1978 **Chemistry** Graham L. Patrick, Wilson Flood, 2002 Keyboard, 2001 **Organic Chemistry NOTE. 3** | 100,2016-12-03 Electronic Musician ,2001 Organic Chemistry Iii Research And Education Science Citation Index ,1992 Vols for 1964 have guides and journal lists **Atomic-Scale Insights** Association, 2002 into Emergent Photovoltaic Absorbers Alex Ganose, 2020-08-29 This book presents an original investigation into alternative photovoltaic absorbers Solar power is a highly promising renewable energy solution however its success is hampered by the limited cost effectiveness of current devices The book assesses the photovoltaic performance of over 20 materials using state of the art first principles methods Adopting a computational approach it investigates atomic scale properties at a level of accuracy that is difficult to achieve using laboratory based experimental techniques Unlike many theoretical studies it provides specific advice to those involved in experimental investigations Further it proposes directions for future research This book advances the field of photovoltaics in three crucial ways firstly it identifies why one class of proposed materials cannot achieve high efficiency while at the same time gaining insights that can be used to design future absorbers Secondly it shows that poor performance in the bismuth chalcohalides is not due to fundamental limitations and can be overcome by finely controlling synthesis conditions Lastly it describes a range of new stable materials that are expected to show excellent photovoltaic performance C342 Lawrence Montgomery, 1998 Lecture Notes on Organic Chemistry August David Burk, 1907 Lecture notes taken by Charles A D Burke Princeton Class of 1909 of lectures given by General Organic Prof Neher dated 1907 1908 Organic Chemistry (Lecture Notes) Nyenty Arrey, 2007-01-01 Chemistry Chm 233 Gapped Lecture Notes Ian Gould, 2016-08-16 **Outline and Notes in Organic Chemistry and Biochemistry** Antonia P. Gabriel, 1967 Hand-book of Chemistry Leopold Gmelin, 1855 Notes on Organic Chemistry Harvey Clayton Brill,1927 Organic Chemistry Kurt D. Donaldson, Jamie M. Scott, 1996-07-01 **Physical Organic** Chemistry 3,1977 **An Overview of Organic Chemistry** Jeanne Kuhler, 2006-01-01 *Organic Chemistry* Cyril Warcup Wood, Leonard Philip Wakefield, Leslie G. Jaeger, Lionel Percy Smith, Sir George Porter, 1962 Chemistry III University of Sydney. School of Chemistry,

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