

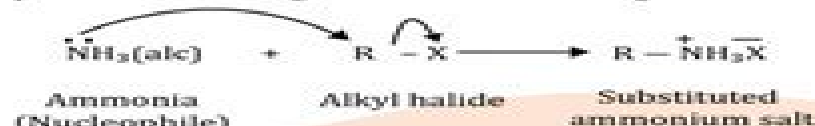
Chemistry

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(Chapter 13) (Amines) (Exercises)

(Class 12)

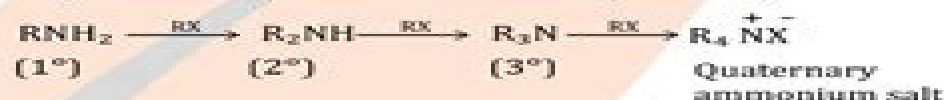
- (v) When an alkyl or benzyl halide is allowed to react with an ethanolic solution of ammonia, it undergoes nucleophilic substitution reaction in which the halogen atom is replaced by an amino ($-\text{NH}_2$) group. This process of cleavage of the carbon-halogen bond is known as ammonolysis.



When this substituted ammonium salt is treated with a strong base such as sodium hydroxide, amine is obtained.



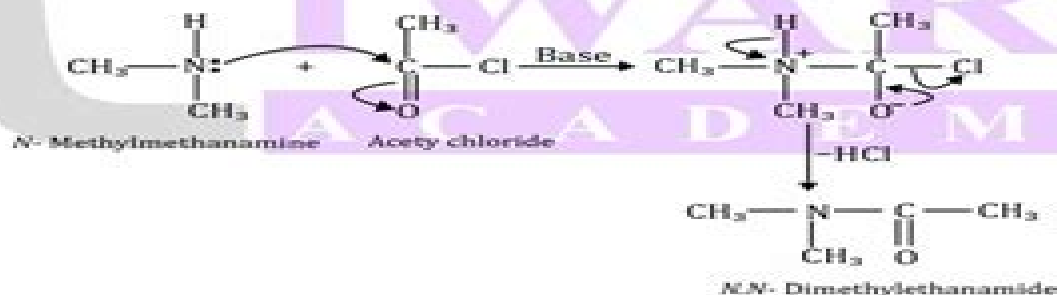
Though primary amine is produced as the major product, this process produces a mixture of primary, secondary and tertiary amines, and also a quaternary ammonium salt as shown.



- (vi) Acetylation (or ethanoylation) is the process of introducing an acetyl group into a molecule.



Aliphatic and aromatic primary and secondary amines undergo acetylation reaction by nucleophilic substitution when treated with acid chlorides, anhydrides or esters. This reaction involves the replacement of the hydrogen atom of $-\text{NH}_2$ or $>\text{NH}$ group by the acetyl group, which in turn leads to the production of amides. To shift the equilibrium to the right hand side, the HCl formed during the reaction is removed as soon as it is formed. This reaction is carried out in the presence of a base (such as pyridine) which is stronger than the amine.



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Table of Contents Ncert 12 Chemistry Guide

1. Understanding the eBook Ncert 12 Chemistry Guide
 - The Rise of Digital Reading Ncert 12 Chemistry Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Ncert 12 Chemistry Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ncert 12 Chemistry Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ncert 12 Chemistry Guide
 - Personalized Recommendations
 - Ncert 12 Chemistry Guide User Reviews and Ratings
 - Ncert 12 Chemistry Guide and Bestseller Lists

5. Accessing Ncert 12 Chemistry Guide Free and Paid eBooks
 - Ncert 12 Chemistry Guide Public Domain eBooks
 - Ncert 12 Chemistry Guide eBook Subscription Services
 - Ncert 12 Chemistry Guide Budget-Friendly Options
6. Navigating Ncert 12 Chemistry Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Ncert 12 Chemistry Guide Compatibility with Devices
 - Ncert 12 Chemistry Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ncert 12 Chemistry Guide
 - Highlighting and Note-Taking Ncert 12 Chemistry Guide
 - Interactive Elements Ncert 12 Chemistry Guide
8. Staying Engaged with Ncert 12 Chemistry Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ncert 12 Chemistry Guide
9. Balancing eBooks and Physical Books Ncert 12 Chemistry Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ncert 12 Chemistry Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ncert 12 Chemistry Guide
 - Setting Reading Goals Ncert 12 Chemistry Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ncert 12 Chemistry Guide
 - Fact-Checking eBook Content of Ncert 12 Chemistry Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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